



# Modelling of hydrogen gas generation from overmature organic matter in the Cooper Basin, Australia

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Andrew Murray<sup>2</sup>, Nicolaj Mahlstedt<sup>3</sup>, Brian Horsfield<sup>3</sup>

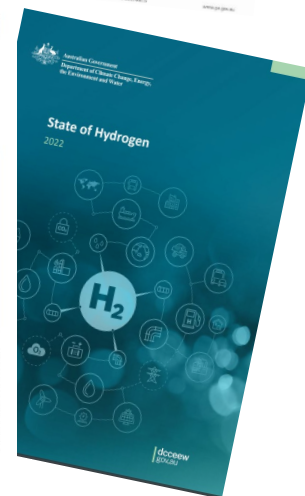
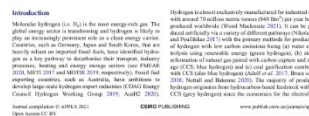
# Exploring for the Future (EFTF) - Low Carbon and Geoscience Advice project

## Mapping H<sub>2</sub> production and storage

- AusH2 – Current hydrogen projects in Australia ([AusH2.ga.gov.au](http://AusH2.ga.gov.au))
- HEFT – Hydrogen Economics Fairways Tool ([ga.gov.au/heft](http://ga.gov.au/heft))
- Searching for salt accumulations in Australia
- Geomechanics of underground hydrogen storage

## H<sub>2</sub> from geologic sources (Natural H<sub>2</sub>)

- Document natural H<sub>2</sub> in natural gases and fluid inclusions
- H<sub>2</sub> surface seepage studies
- Numerical modelling and resource potential of natural H<sub>2</sub>



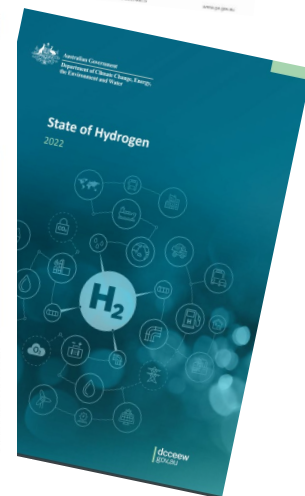
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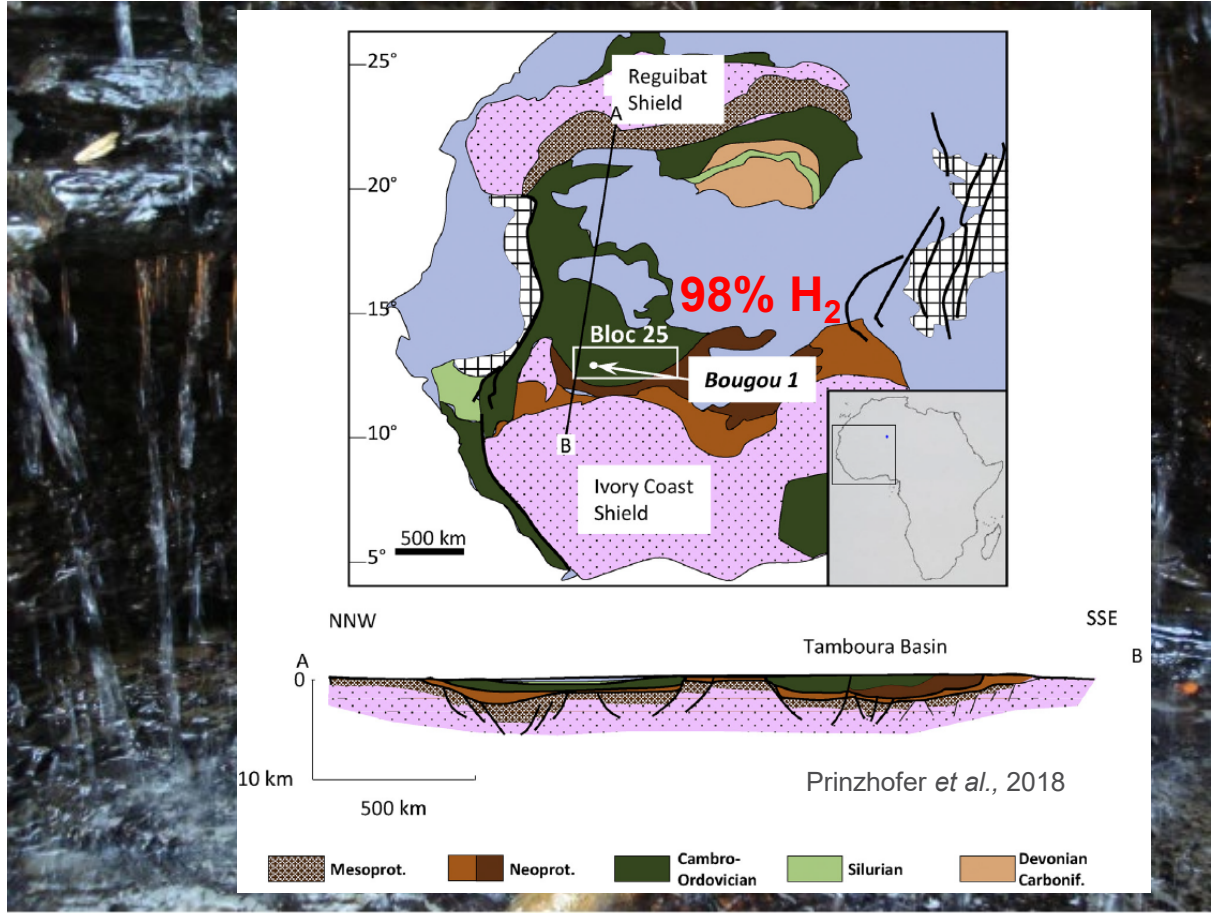
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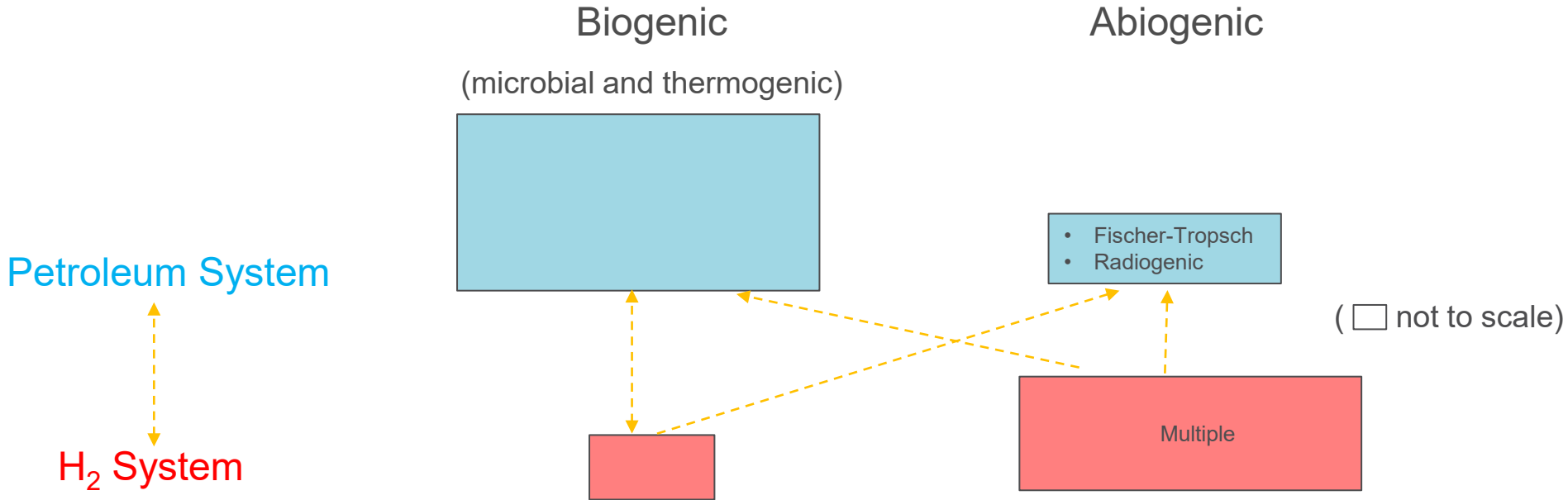


The Chimaera burning gas ( $\text{CH}_4$  and  $\text{H}_2$ ) Turkey.  
 Etiope & Schoell, 2014. DOI: 10.2113/gselements.10.4.291

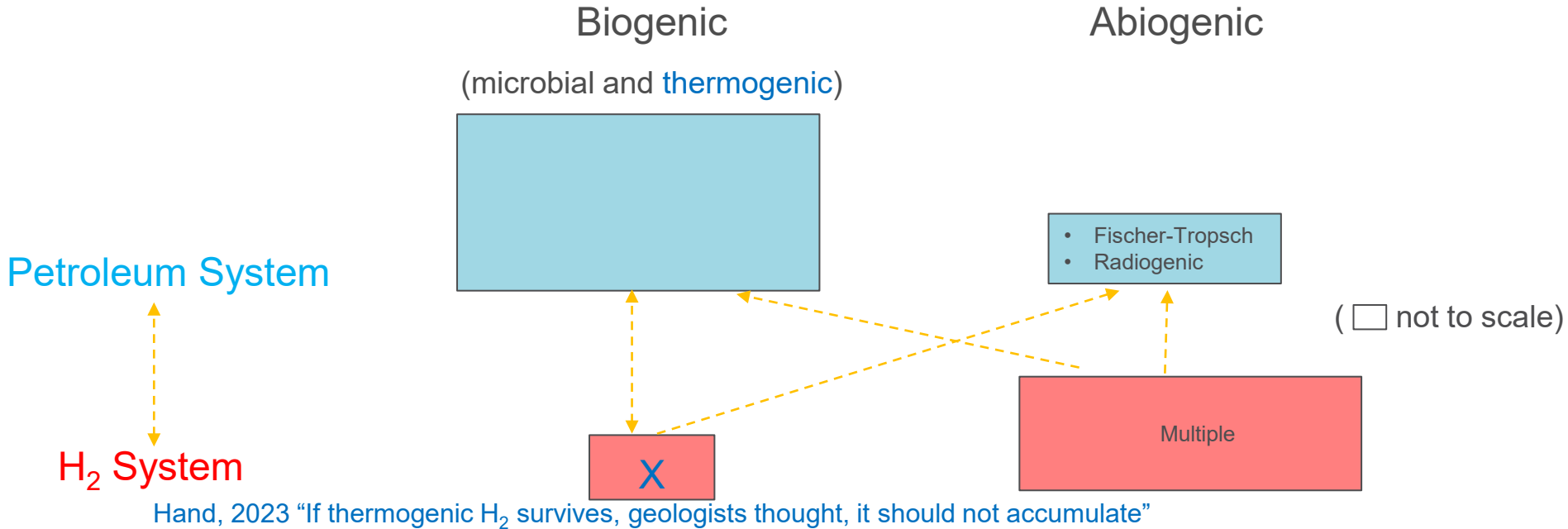


Chesnut Ridge Park, N.Y. <https://www.livescience.com/29510-eternal-flames-natural-gas-source.html>

# Sources of Natural Hydrogen



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## **Serpentinisation of ultramafics ( $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$ )**

Volcanic activity and hydrothermal vents

Cooling of  $\text{CO}_2$ - $\text{CH}_4$ -C fluid systems

Basic magmatic crystallisation

Degassing primordial  $\text{H}_2$

Metasomatisation with metal hydrides

Hydration of biotite

Hydration of siderite

Magnetite to hematite crystallation

Pyritisation

Water hydrolysis

## **Water radiolysis**

Radiolysis of organic matter

Radiolytic dehydrogenation of oil

Radiation induced polymerisation of methane

Cataclastic of silicates

Phosphine hydrolysis

$\text{H}_2$  release from fluid inclusions

Decomposition of methane to graphite

Oxidative coupling of methane

Mixing of water with different ionisation potentials

High temp reaction of ammonium ion with sulfate ion

Dehydrogenation of clay minerals

$\text{H}_2$  from  $\text{H}_2\text{O}$  at high metamorphic temperatures

Biological activity

**Thermogenic OM cracking during aromatisation/condensation**

High temperature decomposition of alkanes & carboxylic acids

Thermochemical sulphate reduction

Drill bit metamorphism

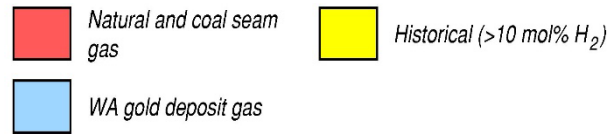
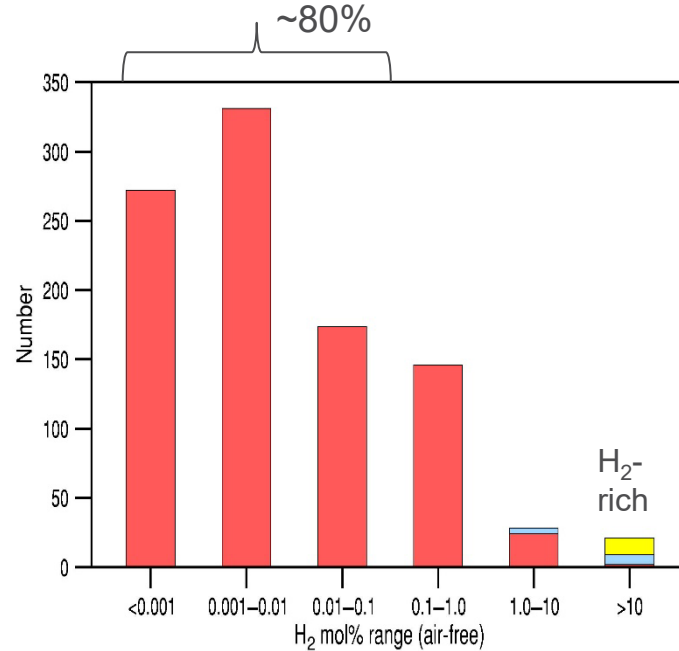
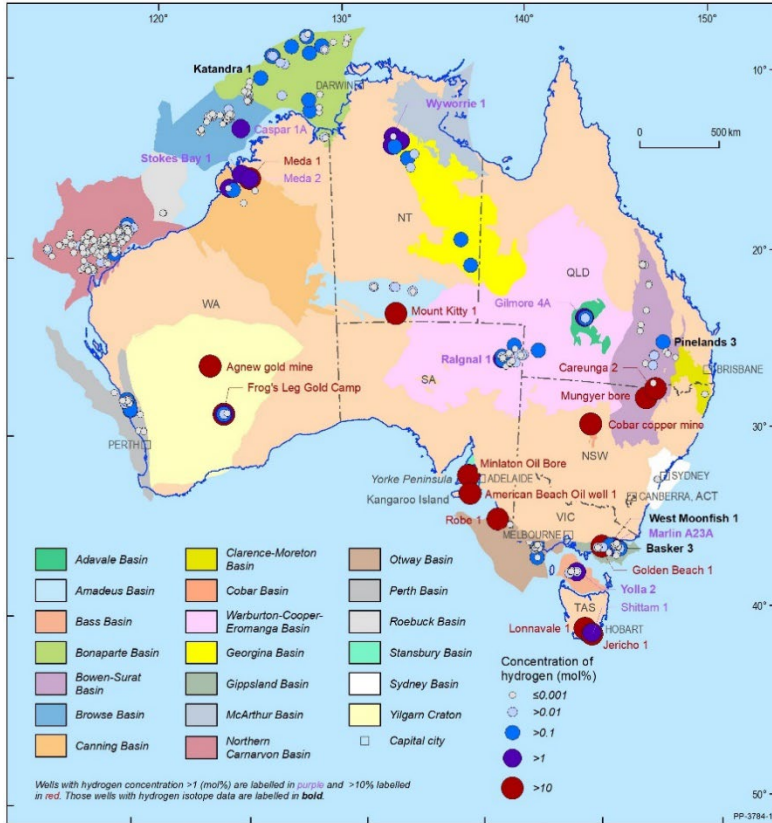
Oxidation of steel pipes

Corrosion of steel well casing by  $\text{H}_2\text{S}$

Acidic corrosion reactions...

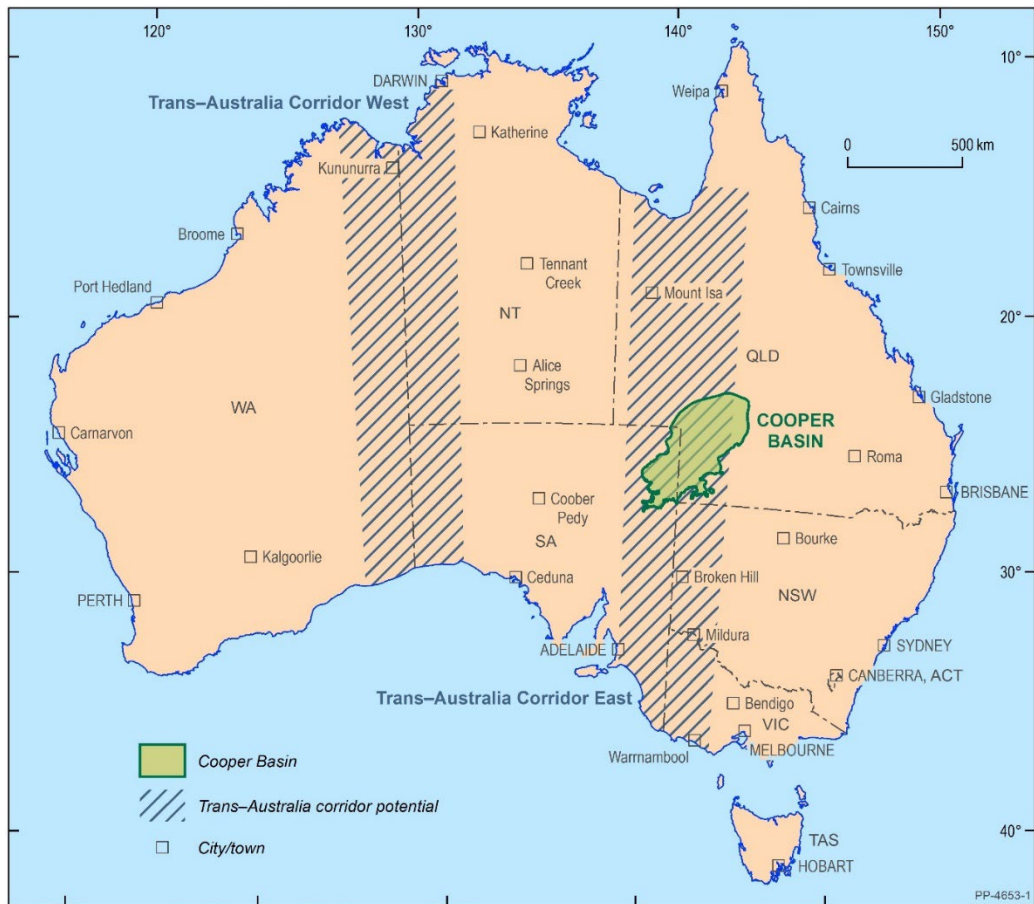
Boreham *et al.*, 2021; Milkov, 2022

# How much and where is hydrogen found in Australia natural gases?

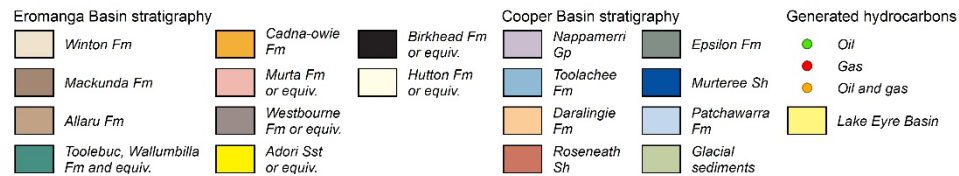
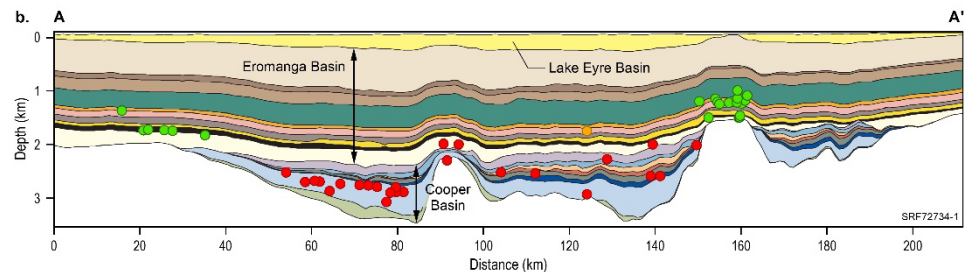
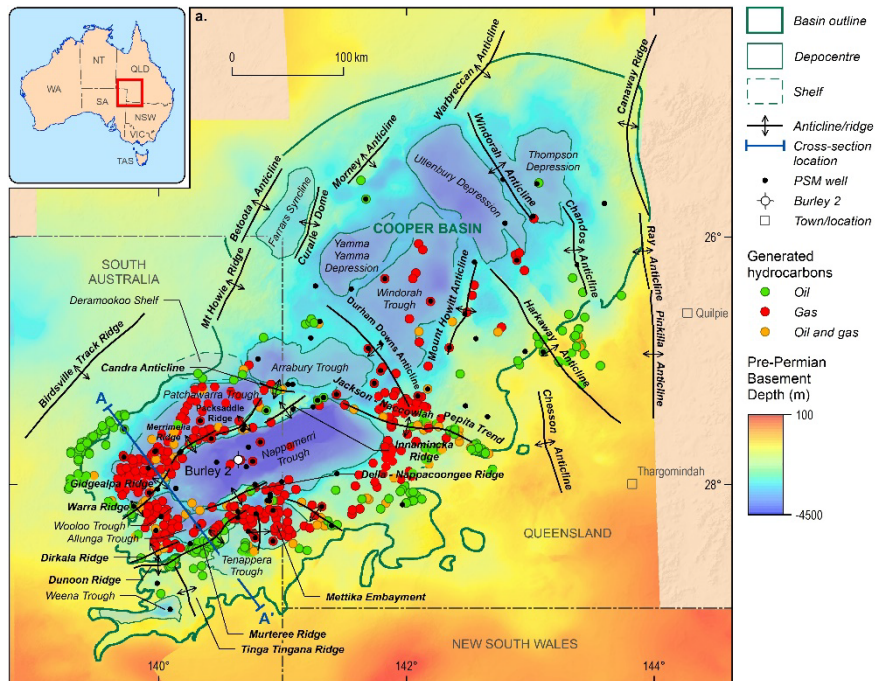


Boreham et al., 2021. APPEA J. <https://www.publish.csiro.au/aj/pdf/AJ20044>



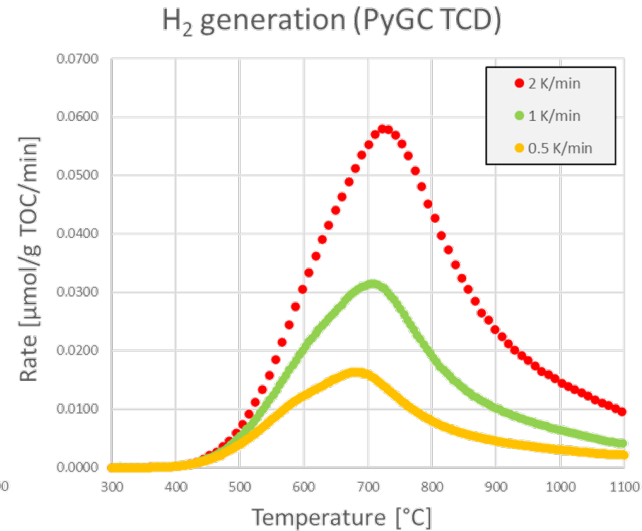
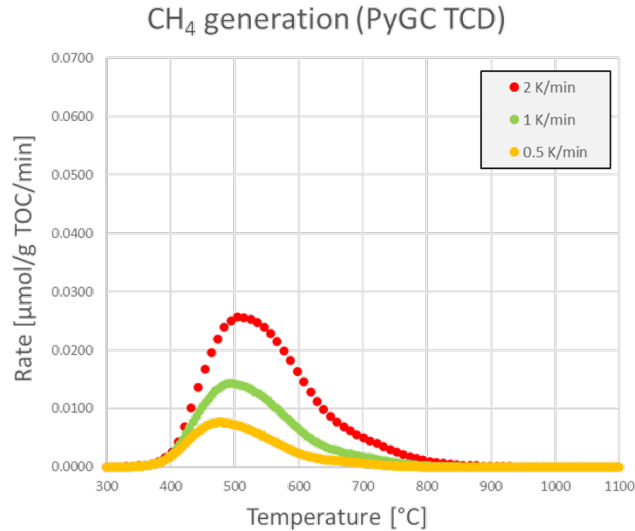
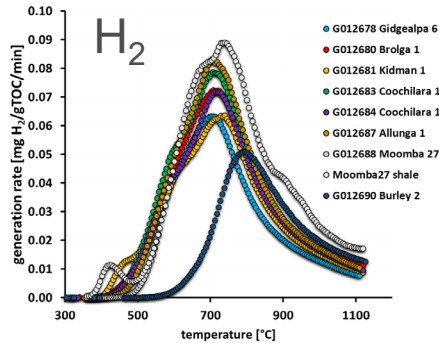
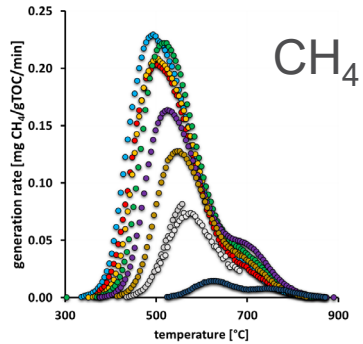


PP-4653-1



Modified from Hall et al., 2019. AAPG Bulletin 103, 31–63. <https://doi.org/10.1306/05111817249>

# Artificial maturation of organic matter (kerogen)



CH<sub>4</sub> and H<sub>2</sub> generation rate curves for marginally mature Gidgealpa-6, Patchawarra Fm. at three different heating rates (0.5, 1.0, 2.0 °C/min).

Open system pyrograms displaying CH<sub>4</sub> and H<sub>2</sub> evolution (mg/gTOC/min) at 1 °C/min for a Patchawarra Fm. maturity sequence

Mahlstedt et. al., 2022. J. Nat. Gas Sci. Eng., 105, 104704

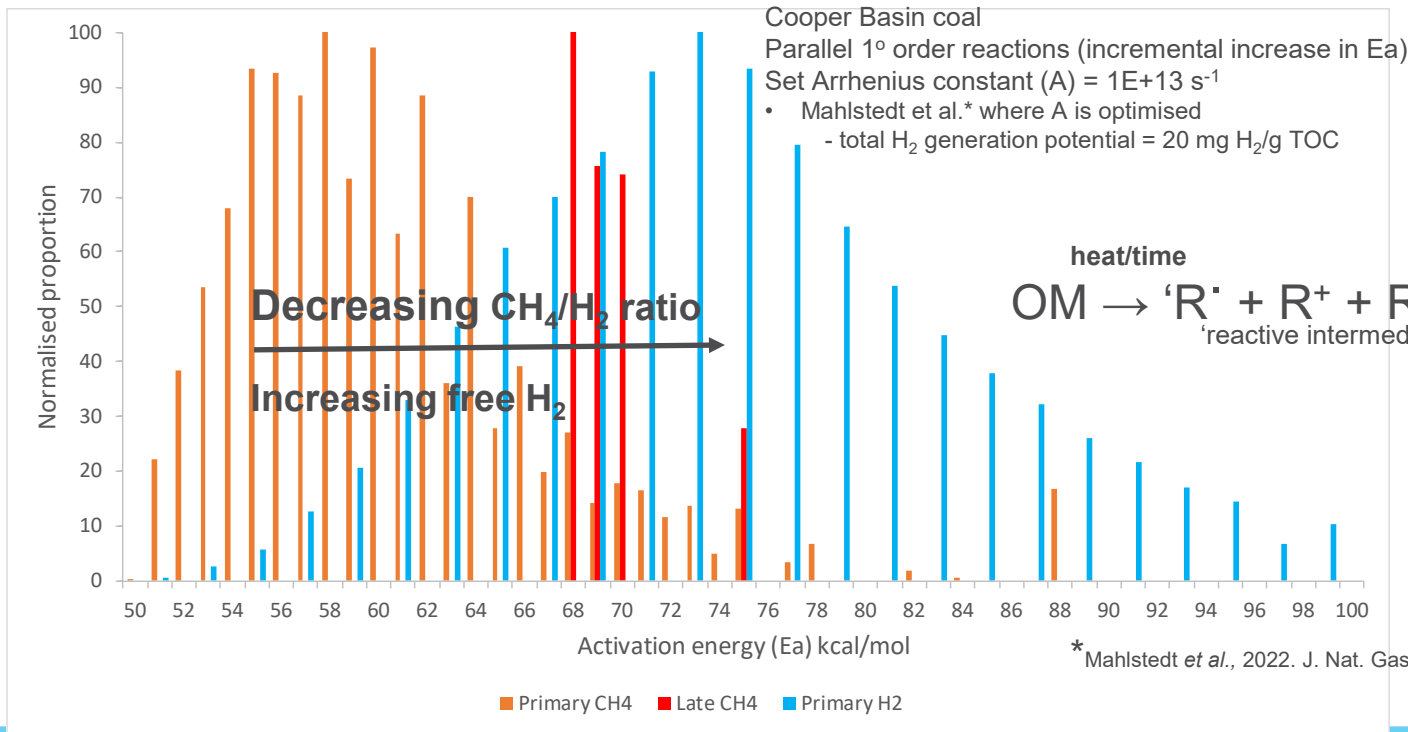
# Chemical kinetics: organic matter (OM) $\xrightarrow{\text{heat/time}}$ H<sub>2</sub> + CH<sub>4</sub> + ...

1<sup>o</sup> order reaction rate of X:  $d[X]/dt = -k[X]$   
 $k_{1..n} = A_{1..n} e^{-E_{a_{1..n}}/RT}$  (Arrhenius approximation)

Primary CH<sub>4</sub>

Late CH<sub>4</sub>

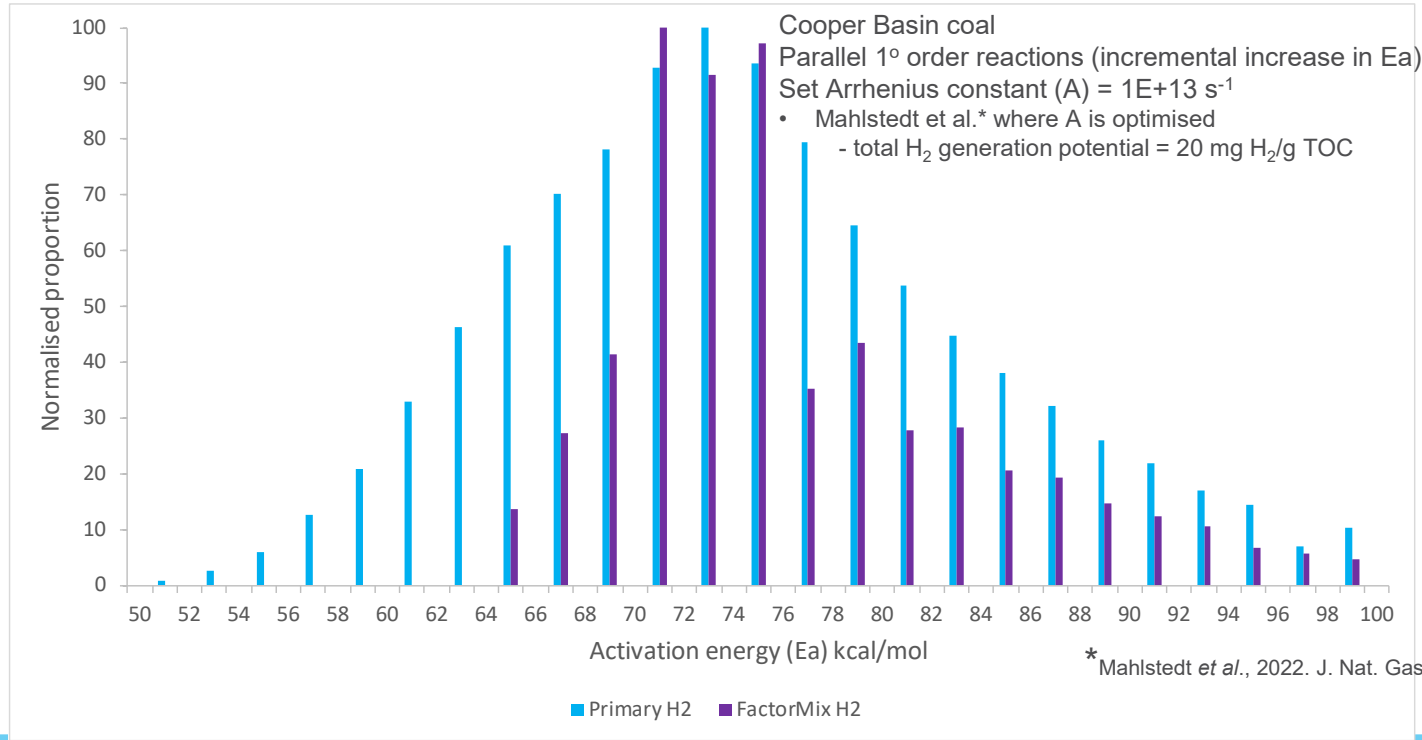
Primary H<sub>2</sub>



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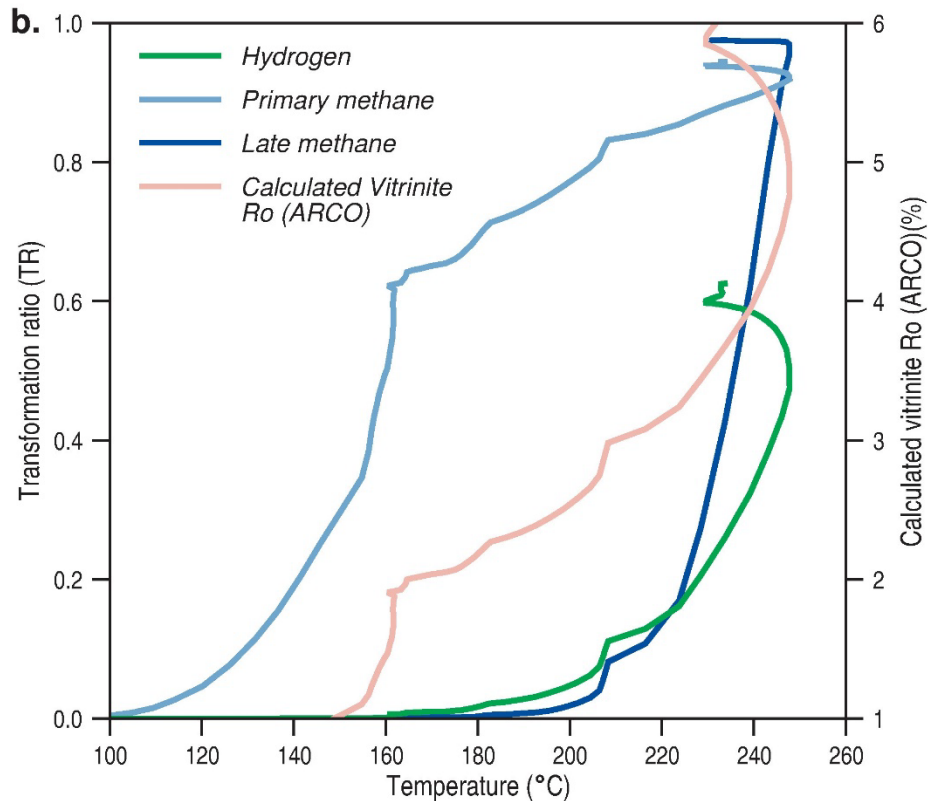
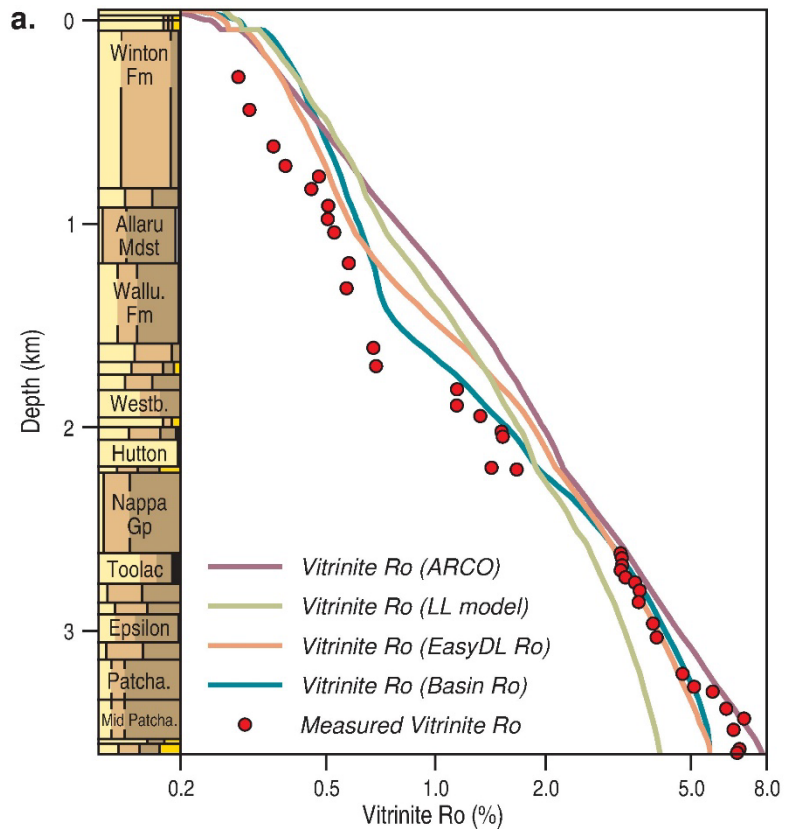
# Chemical kinetics: organic matter (OM) $\xrightarrow{\text{heat/time}}$ $\text{H}_2 + \text{CH}_4 + \dots$

Primary  $\text{H}_2$   
FactorMix  $\text{H}_2$



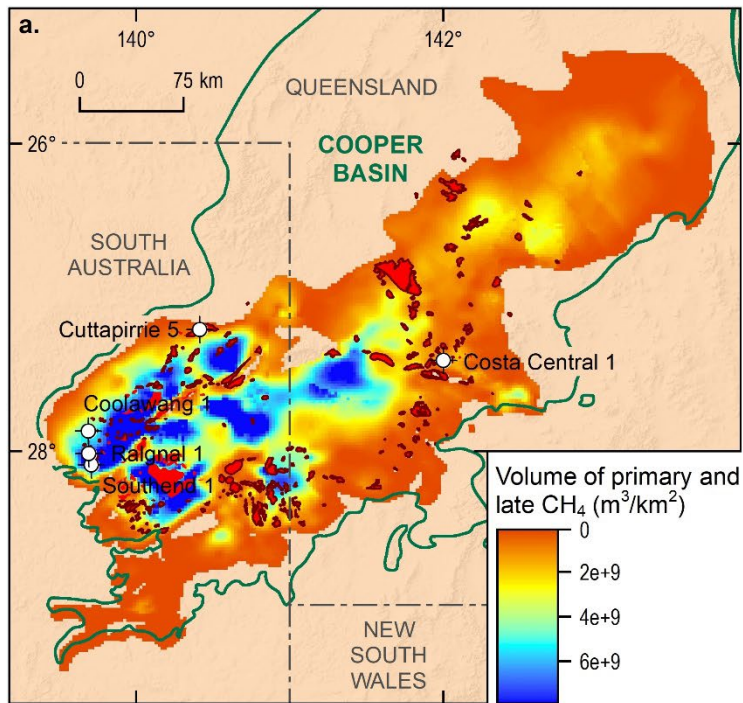
\*Mahlstedt et al., 2022. J. Nat. Gas Sci. Eng., 105, 104704

# Mid-Patchawarra source rock: thermal & maturation history at Burley-2



# Mid-Patchawarra coal and carbonaceous shale

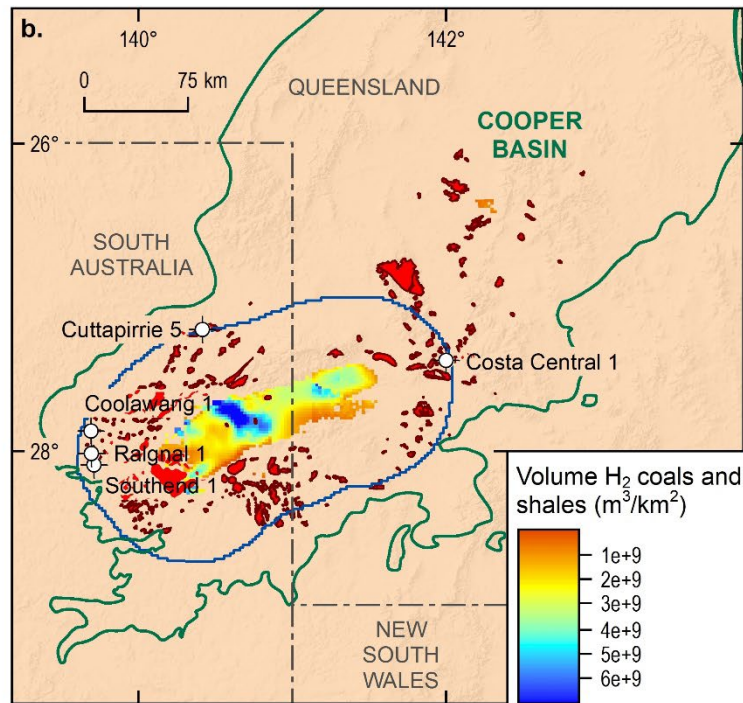
## Primary Methane + Late Methane



Primary methane = max. 50 mg  $\text{CH}_4/\text{g}$  TOC

Late methane = max. 40 mg  $\text{CH}_4/\text{g}$  TOC

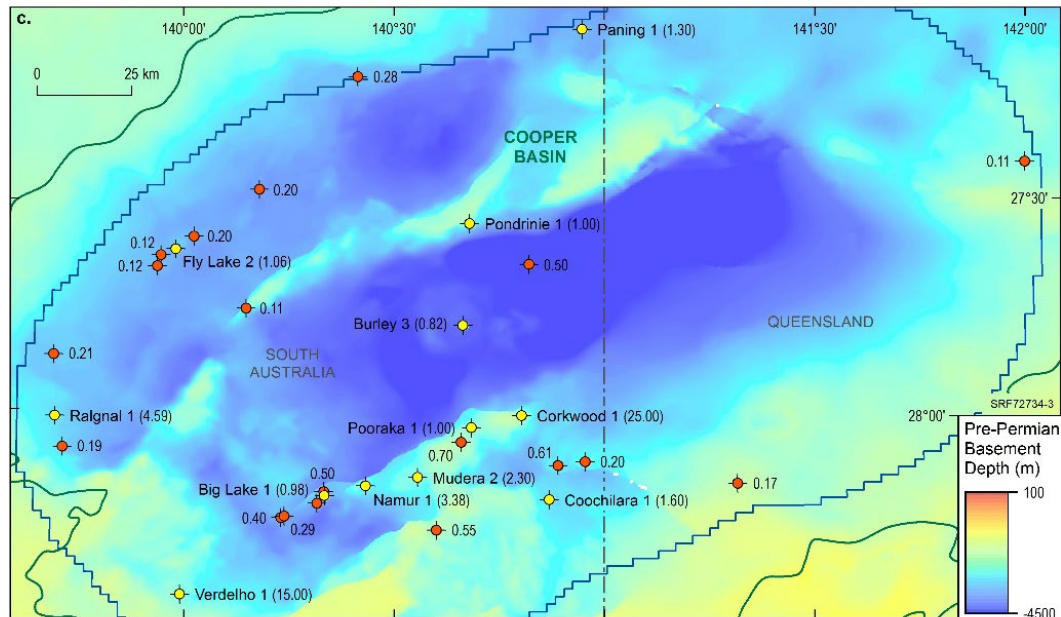
## Thermogenic $\text{H}_2$



Thermogenic  $\text{H}_2$  = max. 20 mg  $\text{H}_2/\text{g}$  TOC

# Available H<sub>2</sub> mol% data

WCRs = 3346 wells  
Wells with gas data = 557  
Wells with H<sub>2</sub> mol% = 35  
(108 analyses)



Wells with H<sub>2</sub> > 0.1 mol% = 26  
Wells with H<sub>2</sub> > 1 mol% = 9

- Cooper Basin
- Gas field
- 50 km boundary from edge of H<sub>2</sub> sweet spot
- Petroleum well
- Wells with H<sub>2</sub> ≥ 0.8 mol%
- Wells with 0.1 mol% < H<sub>2</sub> < 0.8 mol%



# Conclusions

Natural H<sub>2</sub> has multiple sources (including thermogenic) from a wide range of geological settings and rock types → enormous potential resource (large # niche plays).

Cooper Basin is under-explored for (thermogenic) H<sub>2</sub> resource potential.

Natural H<sub>2</sub> has good **Green** credentials:

- Renewable;
- Blended gas transmission pipelines where >10 mol% H<sub>2</sub> (CH<sub>4</sub>, low in CO<sub>2</sub>);
- Where N<sub>2</sub> is high → concentrate H<sub>2</sub> (LNG He spin-off?);
- Use directly (98 mol% H<sub>2</sub>) (e.g., Mali, W. Africa).