

Supplementary Material

Aging of silver nanocolloids in sunlight: particle size has a major influence

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Table S1: AF4 method

Detector flow (mL min⁻¹)	1
Injection flow (mL min⁻¹)	0.2
Focus flow (L min⁻¹)	0.5
Injection time (min)	5
Elution	Constant cross flow 1mL min ⁻¹ during 1 min, then exponential crossflow decrease to 0.1 mL min ⁻¹ for 20 min

Table S2: ICP-MS experimental conditions

RF power (W)	1550
Plasma gas (Ar) flow rate (L min⁻¹)	15
Carrier gas (Ar) flow rate (L min⁻¹)	1.15
Sampling depth (mm)	7.5
Integration time / mass (s)	1

Table S3: Summary of the experimental conditions

	Concentration	Medium	pH
Aging	~20 mg L ⁻¹	Water/citrate stabilizer	7.9 +/- 0.6
UV/Vis	~4 mg L ⁻¹	UP water	6.5 +/- 0.5
DLS	~4 mg L ⁻¹	1 mM NaNO ₃	8.0 +/- 0.5
AF4-MALS-ICP-MS	~1 mg L ⁻¹	NaOH (eluent)	10.0 +/- 0.2
TEM	~20 mg L ⁻¹	Water/citrate stabilizer	7.9 +/- 0.6

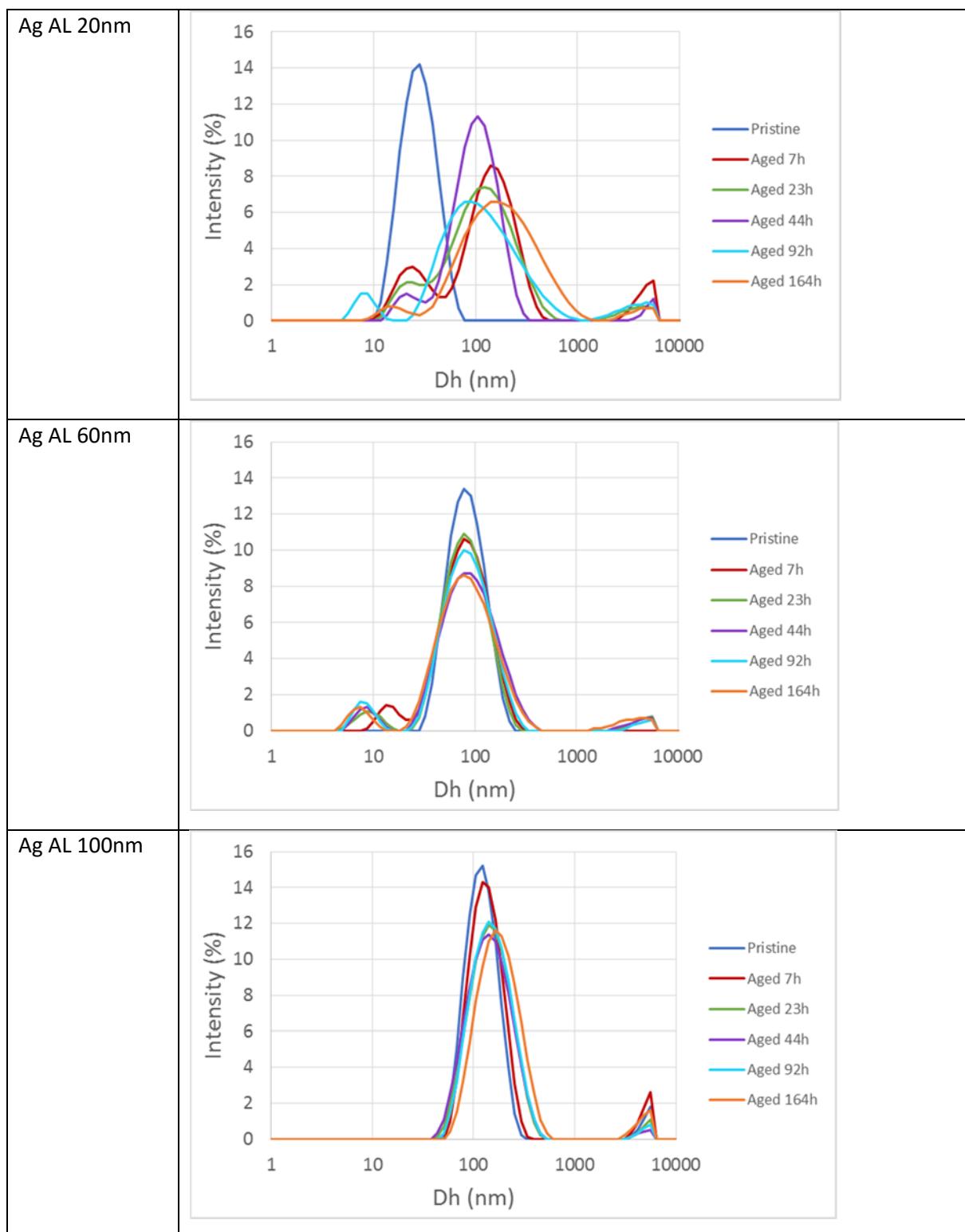
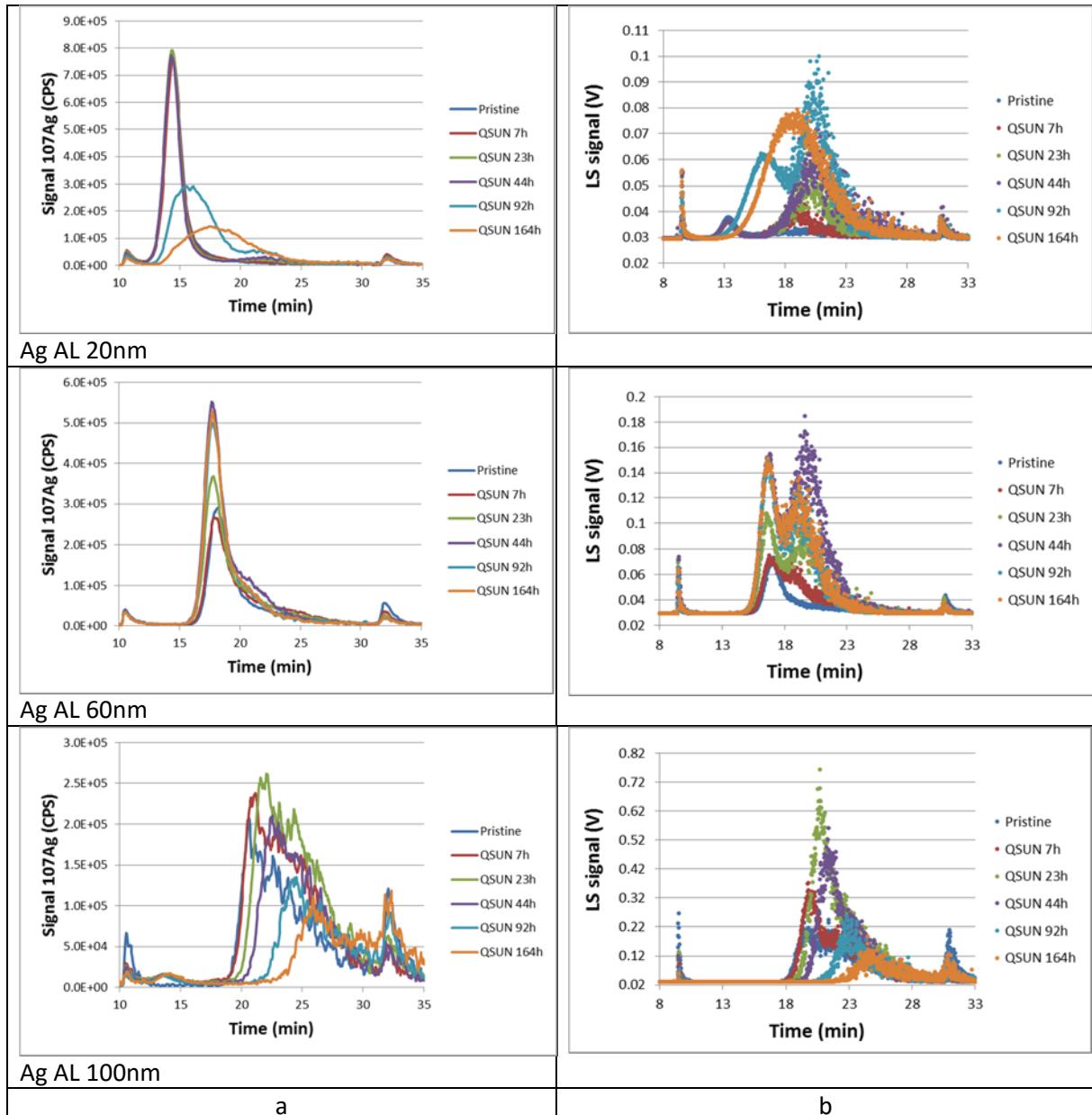
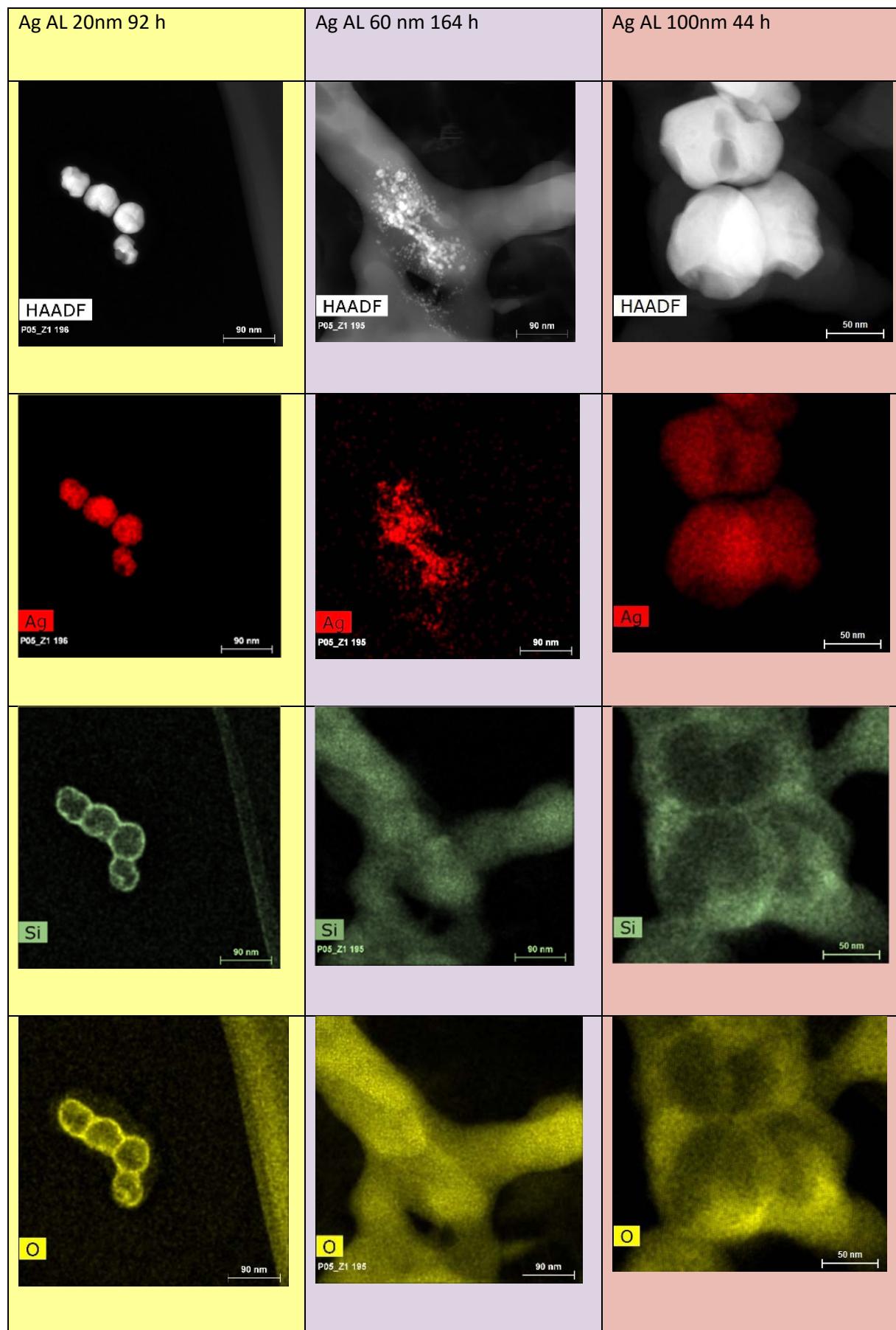


Figure S1: Evolution of the hydrodynamic diameter distribution (DLS, intensity based) of the three populations Ag AL 20nm, AG AL 60nm, and Ag AL 100nm during the aging process. Aging process in QSUN (sunlight, 40°C). Dilution factor: 5 in 1 mM NaNO₃. pH = 8.0 ± 0.5. The suspensions were vortexed for homogenization and sonicated for 1 min to re-disperse the precipitate when necessary (Ag AL 100nm). Repeatability was poor in some instances (particularly, the smallest population was not present in all the replicates) and the choice was made to display the curves showing a maximum number of peaks.





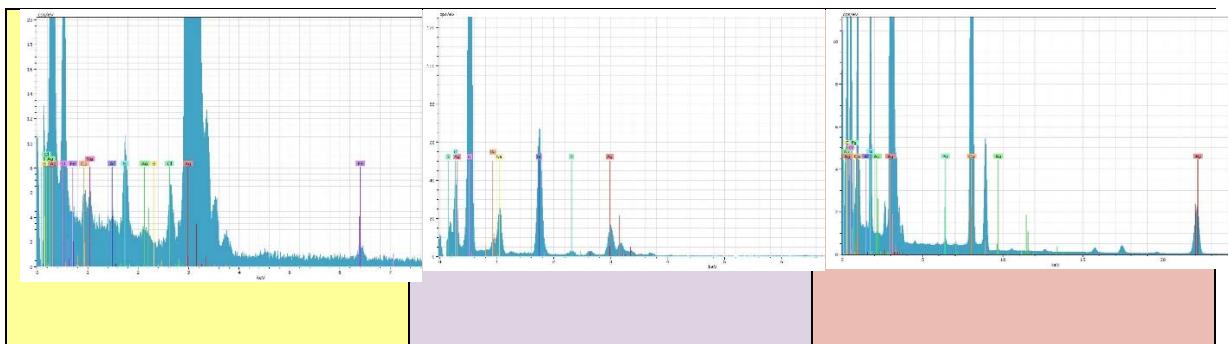


Figure S3: EDS mapping and spectra of some relevant changes in the particle morphology/size/aggregation after aging.