

SUPPLEMENTARY MATERIAL

Application of the Variable Oxygen Probe to Derivatives of 2,6-Dimethyltetrahydropyran-2-ol: Evidence for Through- bond n_O - σ_{CC} - σ^*_{CO} Interactions

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SUPP Figures S1-S16 Thermal Ellipsoid plots for all compounds.

Ellipsoids are at the 50% probability level for all structures.

Those structures with $Z' = 2$ the entire contents of the asymmetric unit are shown. Pages 2-6

SUPP Figure S17 Plot of anti-CC vs pKa (ROH) in **4** Page 7

SUPP Figure S18 Plot of ring CO vs pKa (ROH) in **4** Page 7

SUPP Figure S19 Plot of ring CC(OR) vs pKa (ROH) in **4** Page 8

SUPP Figure S20 Plot of anti-CC vs pKa (ROH) in **5** Page 8

SUPP Figure S21 Plot of ring CO vs pKa (ROH) in **5** Page 9

SUPP Figure S22 Plot of ring CC(OR) vs pKa (ROH) in **5** Page 9

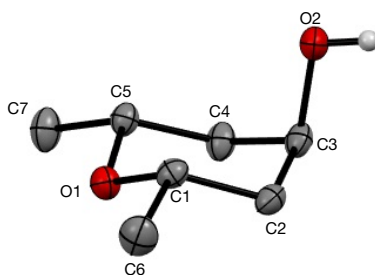
Synthetic procedures and Melting points Page 10-11

¹H NMR Spectra for Synthesized compounds Pages 12-18

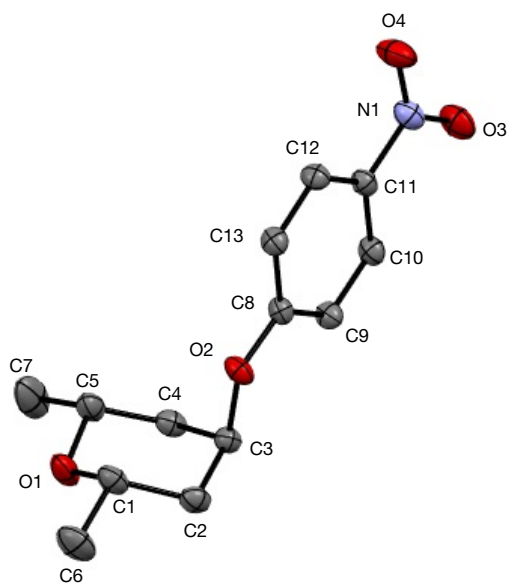
Gaussian Cartesian Coordinates and Energies Page 19-26

Calculated at B3LYP/6-311++G(d,p):

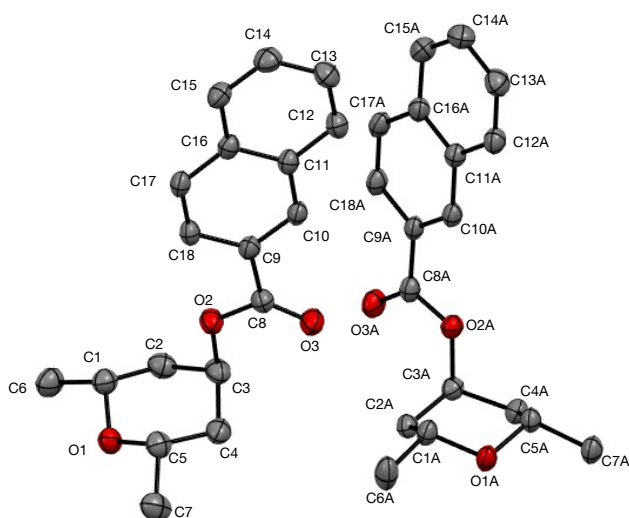
References Page 26



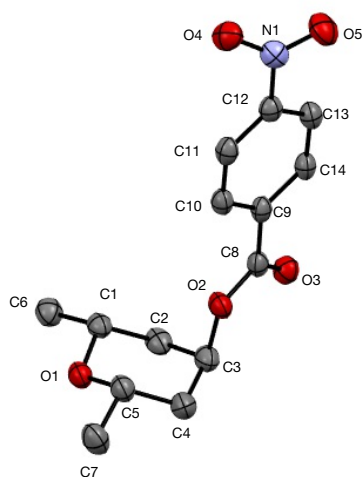
SI Figure S1: Compound 4a.



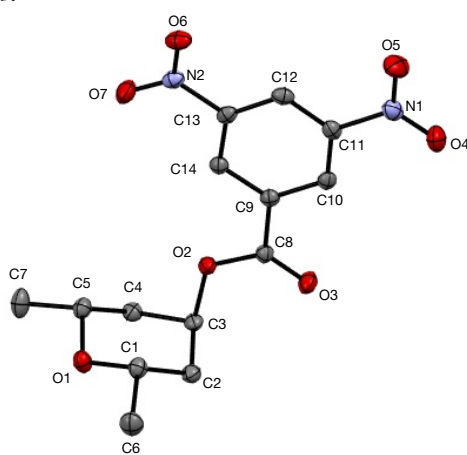
SI Figure S2: Compound 4b.



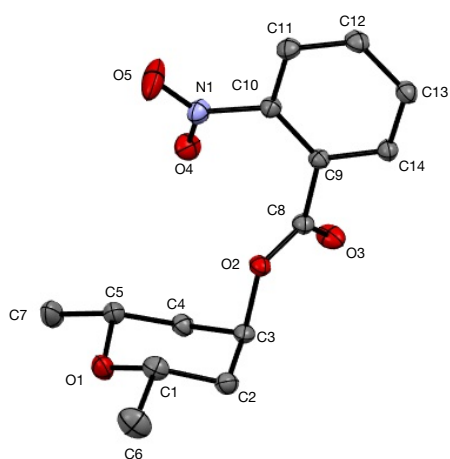
SI Figure S3: Compound 4c.



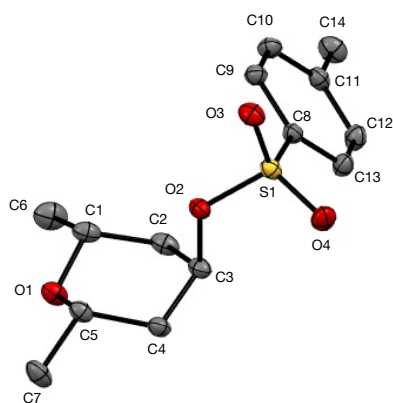
SI Figure S4: Compound 4d.



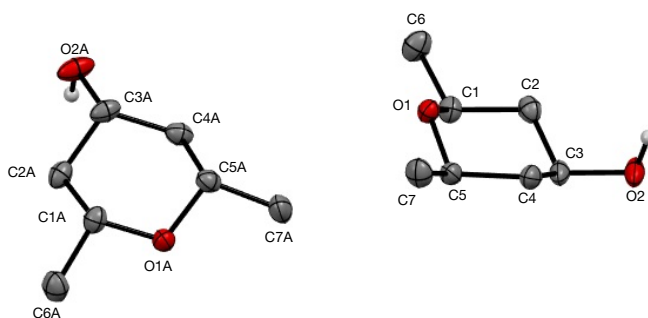
SI Figure S5: Compound 4e.



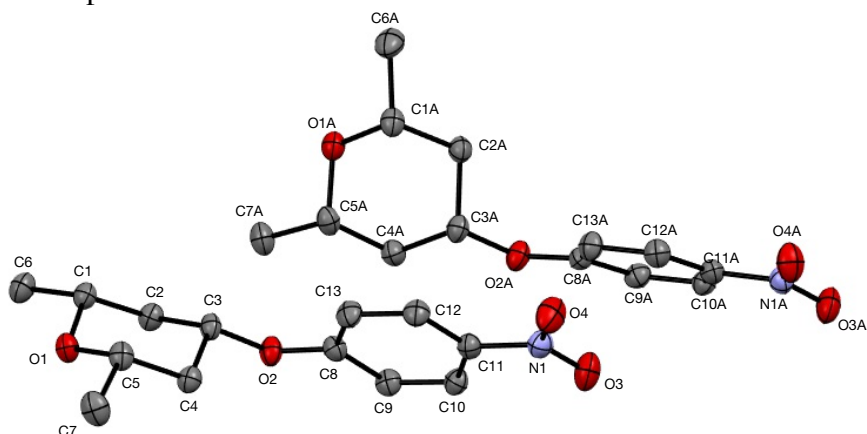
SI Figure S6: Compound 4f.



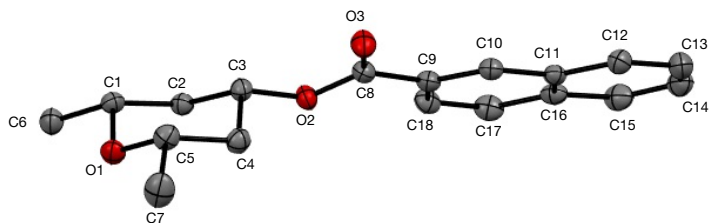
SI Figure S7: Compound 4h.

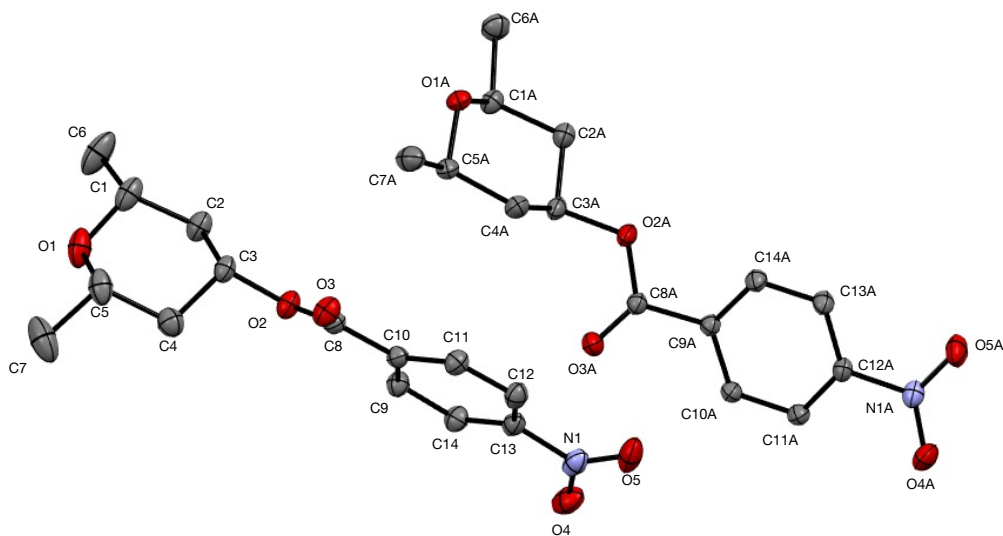
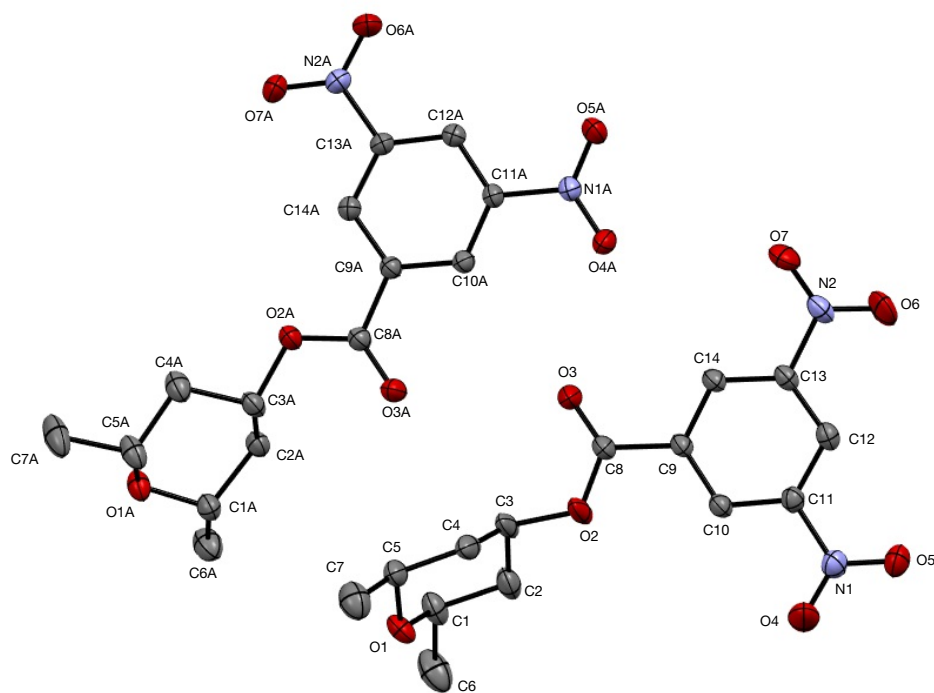
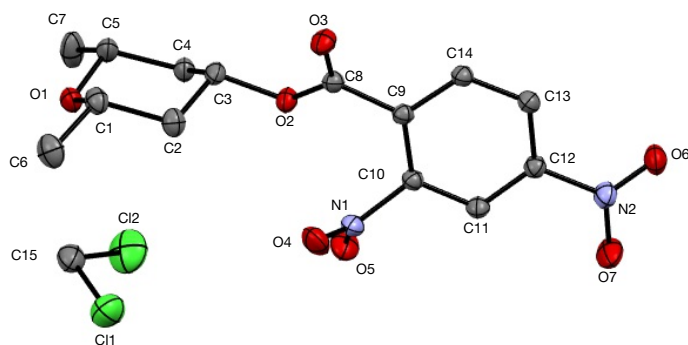


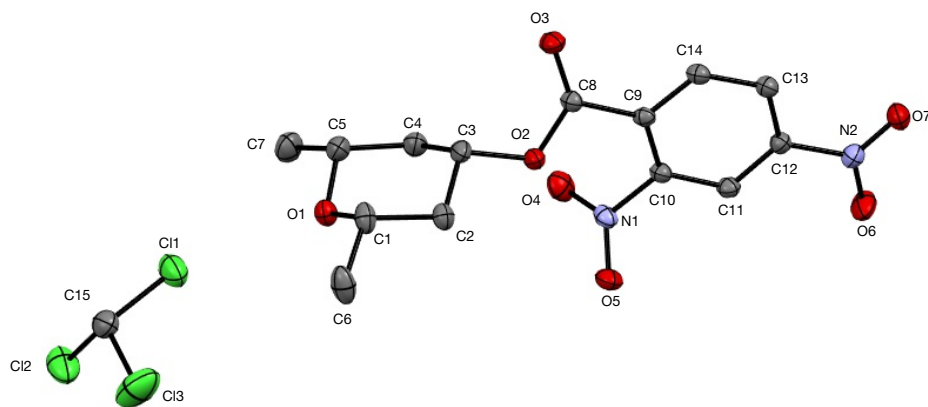
SI Figure S8: Compound 5a.



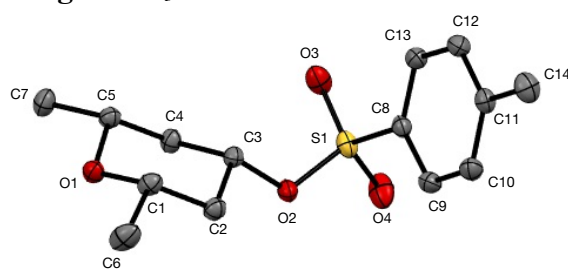
SI Figure S9: Compound 5b.



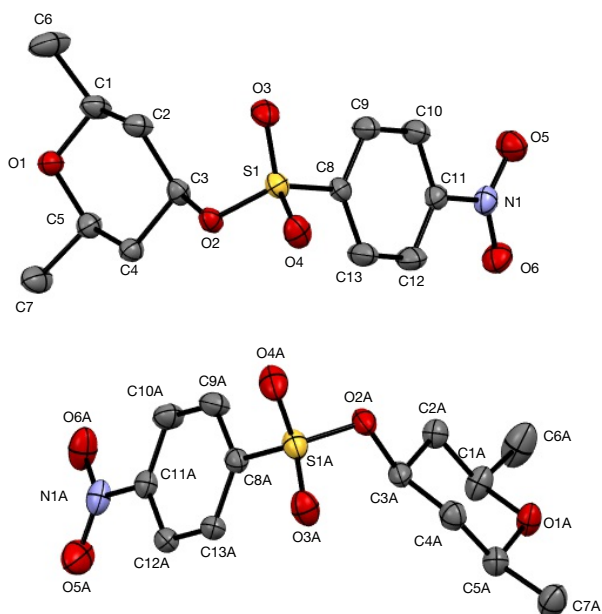
SI Figure S10: Compound **5c**.SI Figure S11: Compound **5d**.SI Figure S12: Compound **5e**.SI Figure S13: Compound **5g**, CH_2Cl_2



SI Figure S14: Compound **5g**. CHCl_3

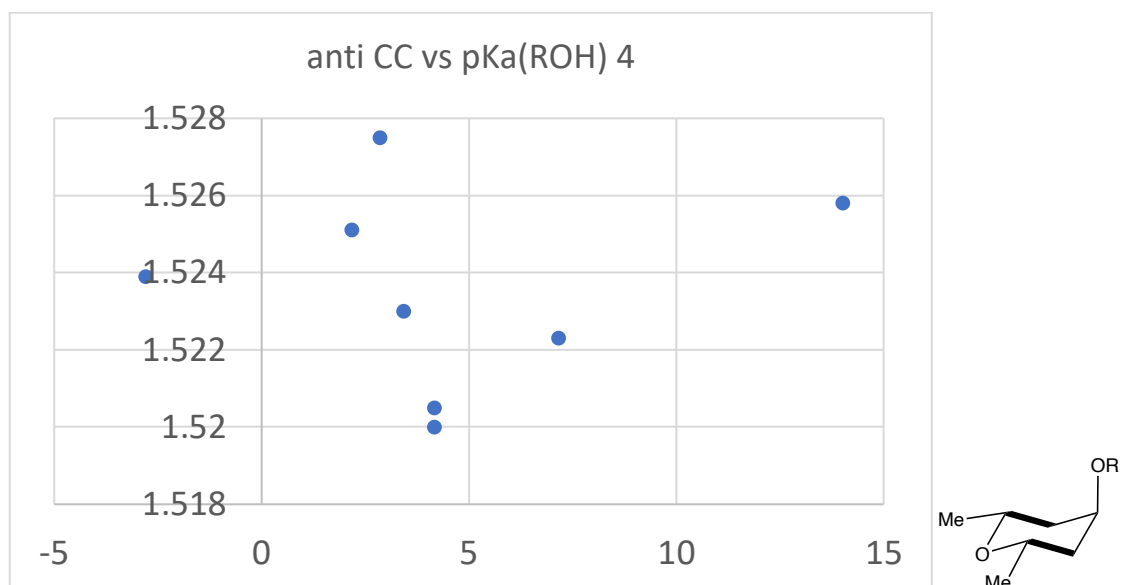


SI Figure S15: Compound **5h**.

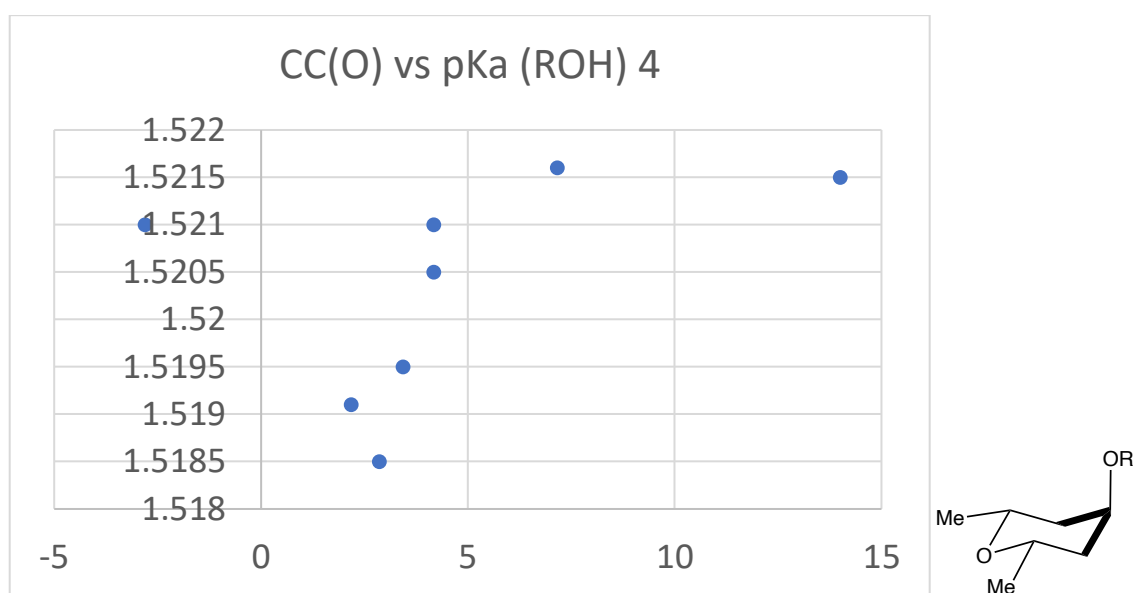


SI Figure S16: Compound **5i**.

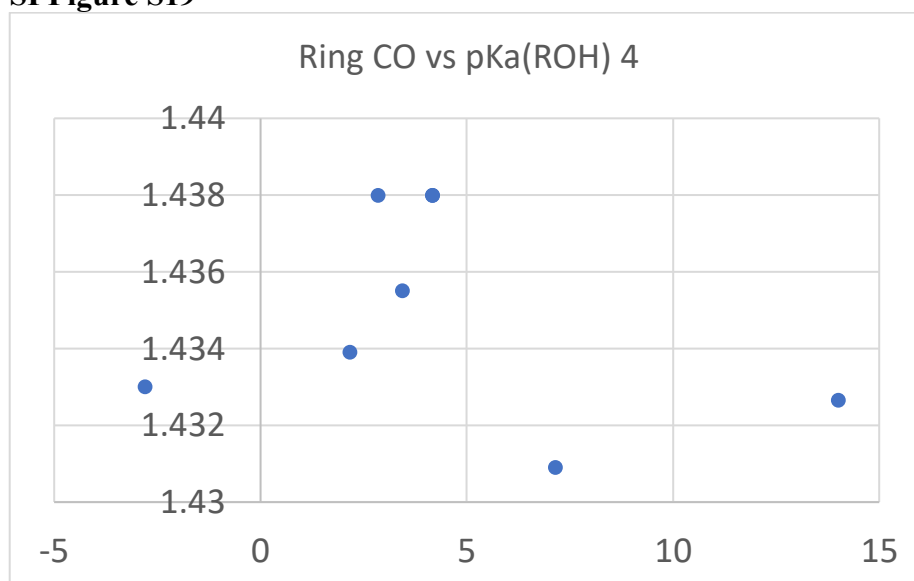
SI Figure S17



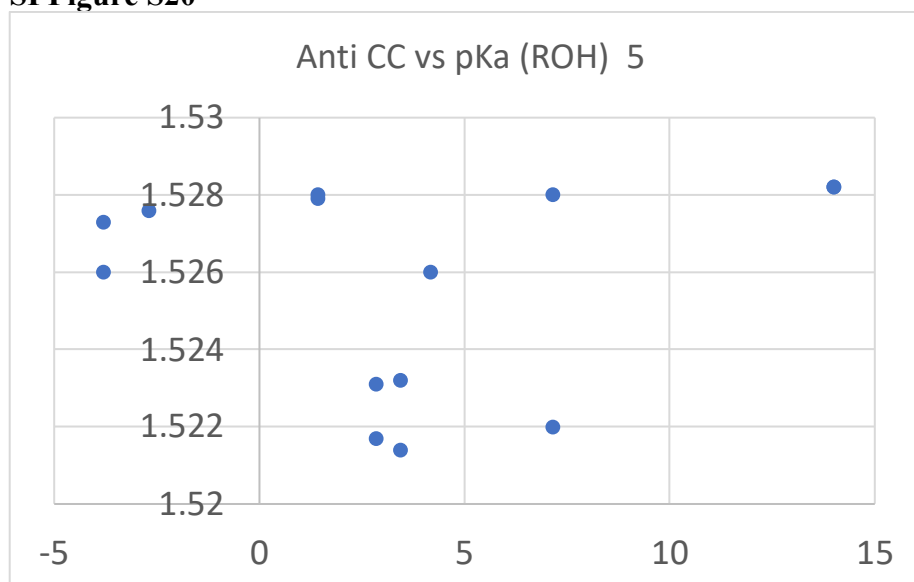
SI Figure S18



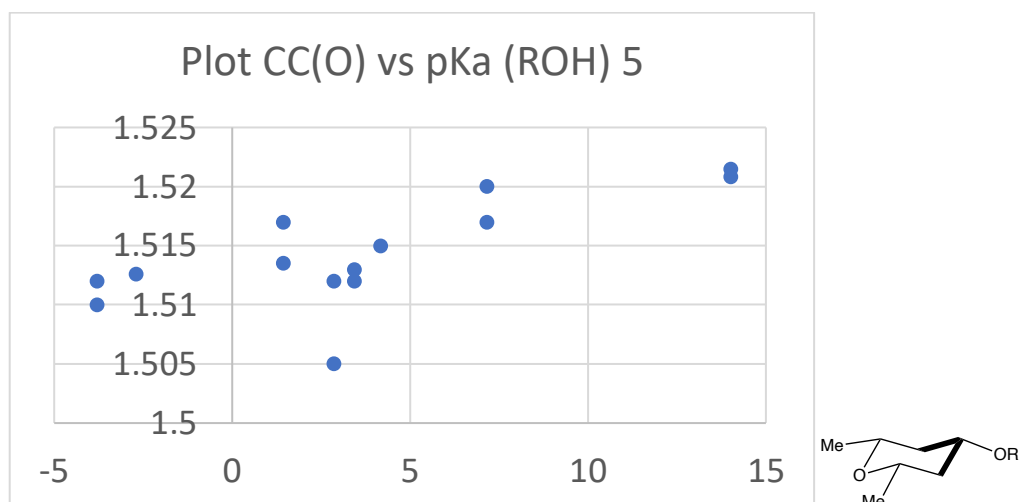
SI Figure S19



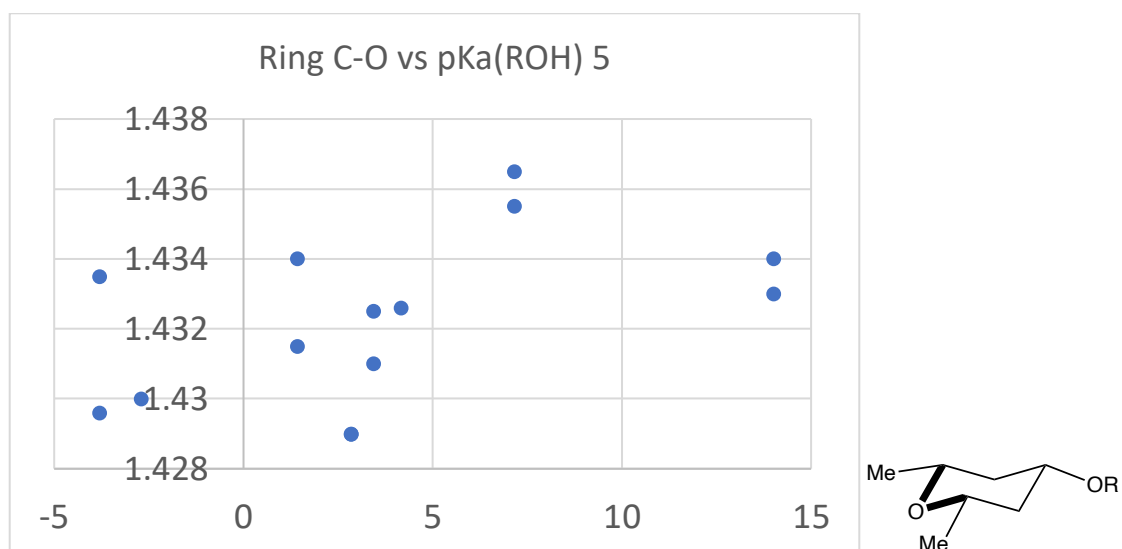
SI Figure S20



SI Figure S21



SI Figure S22



Synthetic procedures.

Procedure for preparation of **4a**, **5a**, **5h**, **4d** and **5d** are adapted from references^{1,2}

Preparation of alcohols **4a** and **5a**

Step 1:

To a solution of homoallylic alcohol (300 mg, 3.48 mmol) in toluene (7.0 mL) at 0 °C were added acetaldehyde (0.097 mL, 1.74 mmol) p-TSA (1.04 g, 5.58 mmol) then the reaction was slowly warm to room temperature and stirred at 70 °C for 3 h. The reaction mixture was diluted with brine solution and extracted with ethyl acetate (2 x 15 mL). The combined organic layers were washed with saturated potassium carbonate solution (15 mL), dried over sodium sulfate and evaporated the solvent to give crude product which was purified by silica gel column chromatography using 20% ethyl acetate-hexanes afforded Prins product (**5h**, 500 mg, 52%). The product was confirmed by ¹H NMR spectroscopy.**Error! Bookmark not defined.**^{Error!}
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Step 2:

To a solution of OTs compound (**5h**, 500 mg, 1.76 mmol) in MeOH (10 mL) was added Mg powder (105.6 mg, 4.40 mmol) at ambient temperature and the resulting mixture was continued to stir at the same temperature for 6 h. After completion of the reaction cooled to 0°C were added conc. HCl (3.0 mL) and water (10.0 mL) and extracted with diethyl ether (3 X 15 mL). The combined organic layers were dried over sodium sulfate, filtered and concentrated under vacuum to give crude product which was purified by silica gel column chromatography using 50% diethyl ether-hexanes obtained desired product (**5a**, 200 mg, 59%). The NMR spectra were identical to literature data.^{1,2}

Step 3:

To a stirred solution of alcohol (**5a**, 200 mg, 1.53 mmol) in anhydrous benzene (6 mL) under argon atmosphere were added benzoic acid (1.15 g, 6.92 mmol) and triphenylphosphine (2.0 g, 7.67 mmol) at room temperature. After 5 min DAID (1.52 mL, 7.67 mmol) was added to the reaction mixture at the same temperature and allowed to continue for 24 h. Then the reaction mixture was concentrated as such (without further work-up) and subjected to silica gel column chromatography using 20% ethyl acetate-hexanes acquired the desired benzoate compound. The product was confirmed by ¹H NMR spectroscopy.

Step 4:

A solution of benzoate above (350 mg, 1.25 mmol) in THF (18 mL) was added LiOH (263 mg, 6.27 mmol) in water (6 mL) at ambient temperature and the resulting mixture was continued to stir at the same temperature for 24 h. THF was removed under reduced pressure and the aqueous layer was diluted with more water (15 mL) and extracted with diethyl ether (3 X 20 mL). The combined organic layers were dried with sodium sulfate, filtered and concentrated under vacuum to give crude product, which was purified by silica gel column chromatography using 50% diethyl ether-hexanes obtained desired product (**4a**, 150 mg, 92%). The NMR spectra were identical to literature data.**Error! Bookmark not defined.****Error!**
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General procedure to prepare benzoates and sulfonates:

To a stirred solution of alcohol (**4a** or **5a**, 100 mg, 0.76 mmol) in pyridine (5 mL) was added acid chloride (293 mg, 1.53 mmol) at 0 °C. Then the reaction was continued to stir for 24 h. Pyridine was removed from the reaction then diluted with water (10 mL) and extracted with ethyl acetate (2 x 10 mL), then the organic layer was washed with aq. HCl (2 x 10 mL). The organic layer was dried over sodium sulfate, filtered, concentrated under reduced pressure and purified by silica gel column chromatography using 5% ethyl acetate-hexanes gave desired product (**4c**, 120 mg, 55%). The products were confirmed by ¹H NMR spectroscopy.

Melting points: **4c** 85-86.4°C, **5c** 101-107°C, **4d** 103-104.5°C, **5d** 93-95.3°C, **4e** 181-182°C, **5e** 99.3-101.6°C, **4f** 62.2-63.7°C, **4h** 60.2-61.3°C, **5h** 85.2-87°C, **4i** 75.6-79.9°C.

General procedure to prepare p-nitrophenoxyl ethers

The alcohol **4a** or **5a** (100 mg) in dry ether (20 ml) was treated with excess potassium hydride (50 mg) and stirred at 0°C for 30 minutes before treatment with p-nitrofluorobenzene (1.05 eq.). After 2 hours the reaction was quenched carefully with water. Excess water was added, and the mixture extracted with ether (3 x 20 ml), dried (MgSO₄), and evaporated down to a residue.

Compound **4b**

¹H NMR (CDCl₃, 400MHz): δ = 8.16 - 8.21 (m, 2 H) (Ar-H), 6.91 - 6.96 (m, 2 H) (Ar-H), 4.82 (bs 1H) (CH_{eq}-OR), 3.87 - 3.96 (m, 2 H) (CO-CH_{ax}-CR), 1.91 (dd, *J*=14.3, 2.7 Hz, 2 H) (RC-CH_{eq}-CR), 1.52 (ddd, *J*=14.3, 11.5, 2.7 Hz, 2 H) (RC-CH_{ax}-CR), 1.18 ppm (d, *J* = 6Hz 6H) (CH₃) ¹³C NMR (CDCl₃, 101MHz): δ = 162.5, 141.3, 126.0, 115.2, 71.5, 67.9, 36.4, 31.9, 29.7, 21.8, 14.1 ppm

Compound **5b**

¹H NMR (CDCl₃, 400MHz): δ = 8.12 - 8.17 (m, 2 H) (Ar-H), 6.90 - 6.95 (m, 2 H) (Ar-H), 4.47 - 4.56 (m, 1 H) (CH_{ax}-OR), 3.56 (ddd, *J*=11.4, 6.0, 1.4 Hz, 2 H) (CO-CH_{ax}-CR), 2.05 - 2.12 (m, 2 H) (RC-CH_{eq}-CR), 1.29 - 1.40 (m, 2 H) (RC-CH_{ax}-CR), 1.24 ppm (d, *J*=6.0 Hz, 6 H) (CH₃) ¹³C NMR (CDCl₃, 101MHz): δ = 206.9, 162.6, 141.3, 125.9, 115.3, 74.2, 71.3, 38.7, 30.9, 29.7, 21.7 ppm

Compound **4f**

¹H NMR (CDCl₃, 400MHz): δ = 7.91 (d, *J*=8.2 Hz, 1 H) (Ar-H), 7.59 - 7.74 (m, 3 H) (Ar-H), 5.09 - 5.20 (m, 1 H) (CH_{eq}-OR), 3.51 - 3.62 (m, 2 H) (CO-CH_{ax}-CR), 2.08 (dd, *J*=13.1, 4.5 Hz, 2 H) (RC-CH_{eq}-CR), 1.27 - 1.38 (m, 3 H) (RC-CH_{ax}-CR), 1.22 - 1.26 ppm (d *J* = 6Hz, 6 H) (CH₃) ¹³C NMR (CDCl₃, 101MHz): δ = 164.9, 148.1, 132.9, 131.6, 129.8, 127.9, 123.9, 72.9, 71.3, 59.5, 38.1, 31.2, 29.7, 21.7 ppm

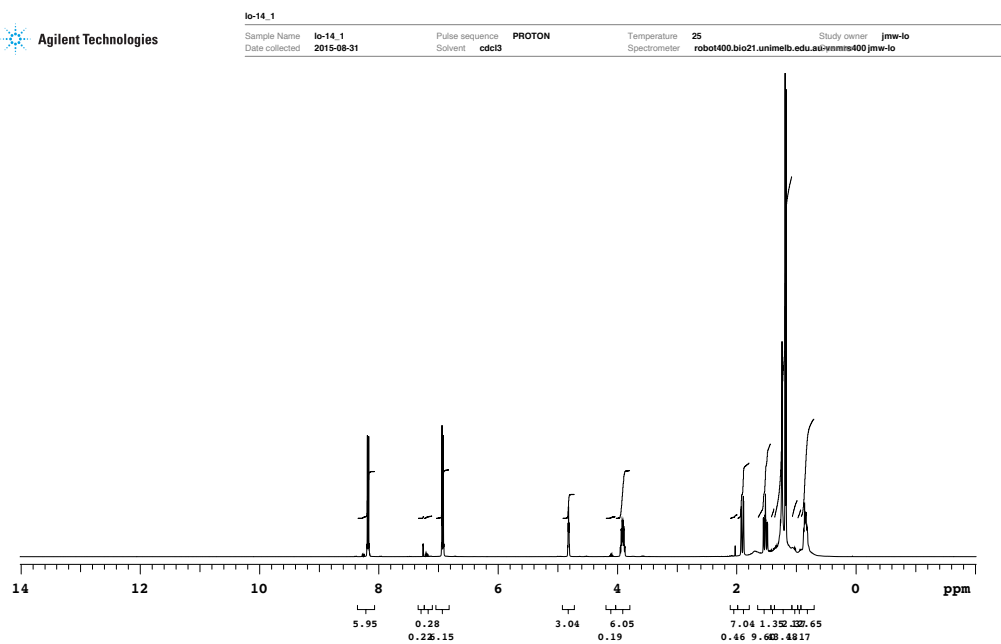
Compound **4g**

¹H NMR (CDCl₃, 400MHz): δ = 8.77 (d, *J*=2.3 Hz, 1 H) (Ar-H), 8.51 - 8.56 (m, 1 H) (Ar-H), 7.98 (d, *J*=8.6 Hz, 1 H) (Ar-H), 5.46 - 5.51 (m, 1 H) (CH_{eq}-OR), 3.67 - 3.77 (m, 2 H) (CO-CH_{ax}-CR), 1.91 (dd, *J*=14.9, 2.3 Hz, 2 H) (RC-CH_{eq}-CR), 1.52 - 1.60 (m, 2 H) (RC-CH_{ax}-CR), 1.17 - 1.22 ppm (d *J* = 6Hz, 6 H) (CH₃) ¹³C NMR (CDCl₃, 101MHz): δ = 162.9, 132.9, 131.4, 127.3, 124.4, 119.5, 72.4, 68.3, 36.5, 31.4, 29.7, 21.8 ppm

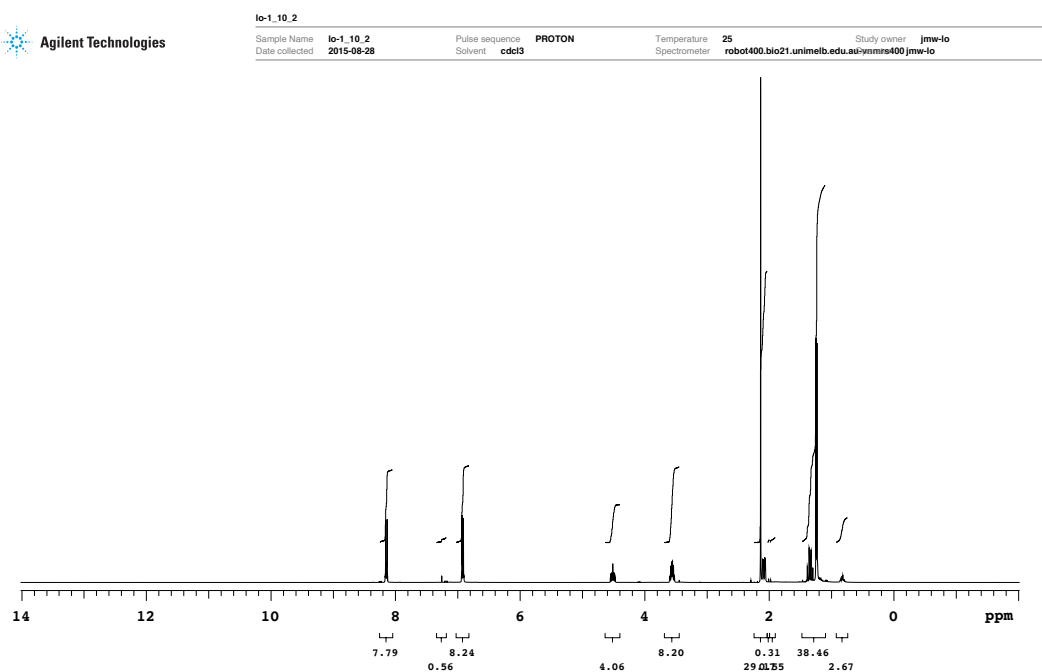
Compound **5g** 78%

¹H NMR (CDCl₃, 400MHz): δ = 8.76 - 8.79 (m, 1 H) (Ar-H), 8.52 (dt, *J*=8.2, 1.8 Hz, 1 H) (Ar-H), 7.91 (dd, *J*=8.2, 1.2 Hz, 1 H) (Ar-H), 5.13 - 5.23 (m, 1 H) (CH_{ax}-OR), 3.51 - 3.62 (m, 2 H) (CO-CH_{ax}-CR), 2.05 - 2.12 (m, 2 H) (RC-CH_{eq}-CR), 1.27 - 1.38 (m, 2 H) (RC-CH_{ax}-CR), 1.21 - 1.26 ppm (d, *J* = 6Hz, 6 H) (CH₃) ¹³C NMR (CDCl₃, 101MHz): δ = 163.2, 148.9, 148.0, 133.1, 131.2, 127.5, 119.6, 74.0, 71.3, 38.0, 21.7 ppm

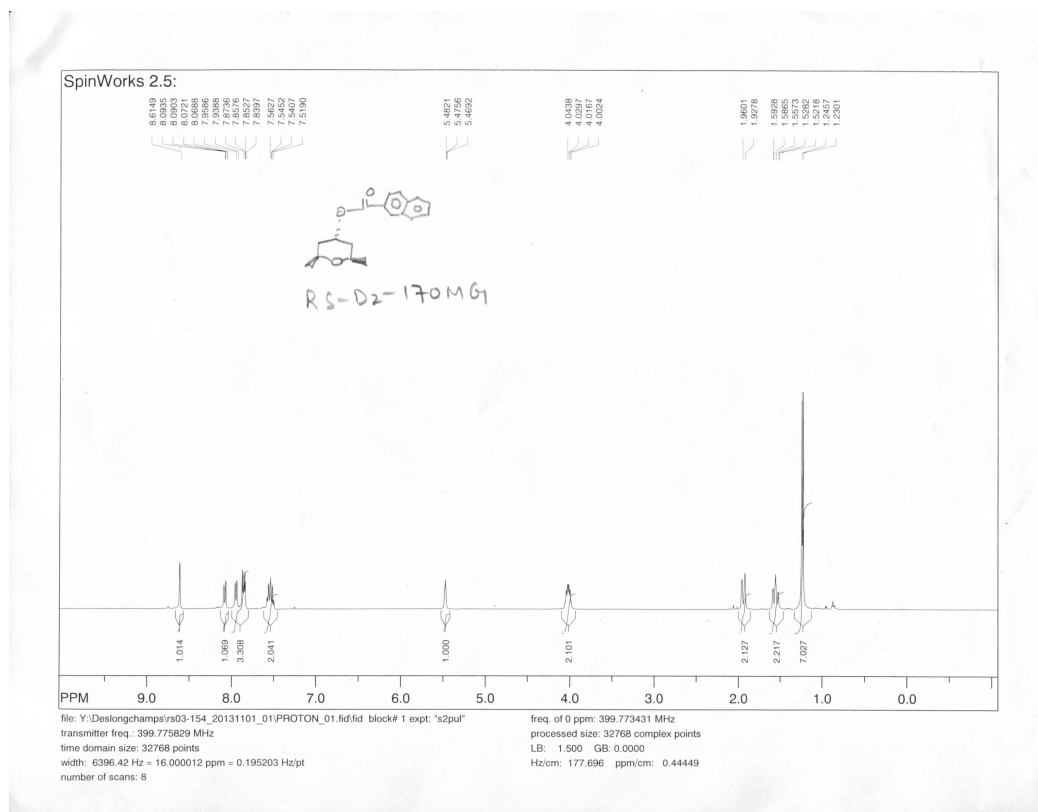
Compound 4b



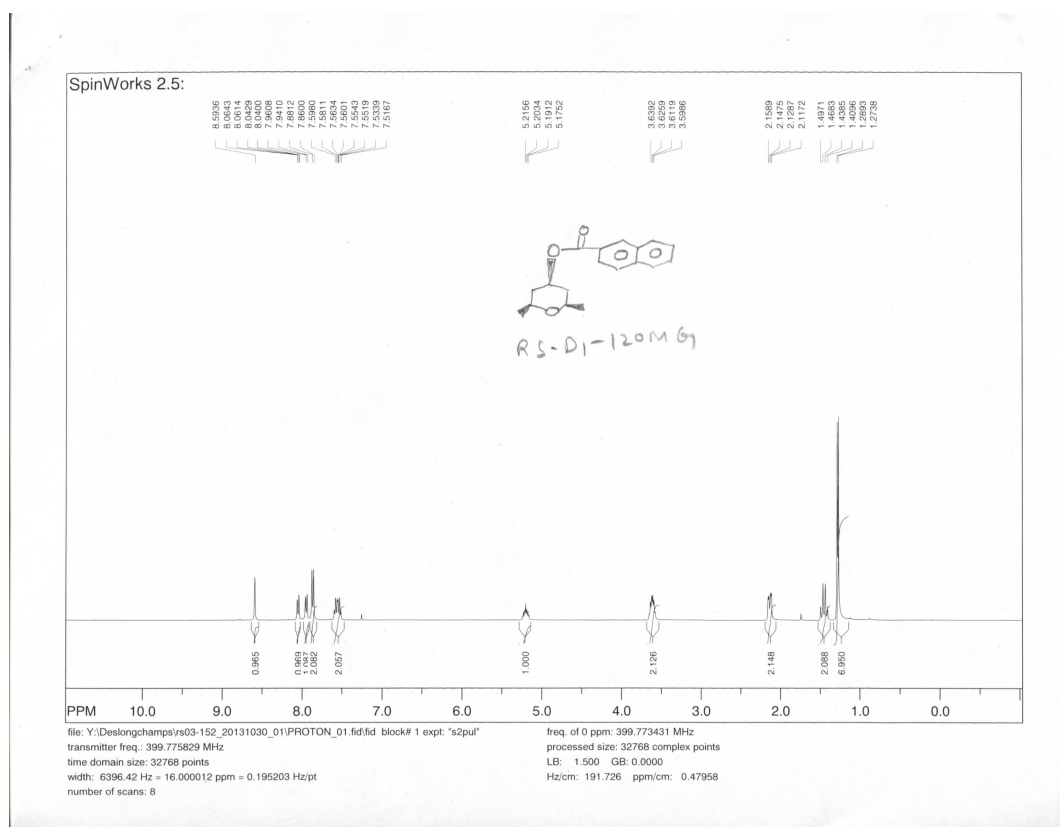
Compound 5b



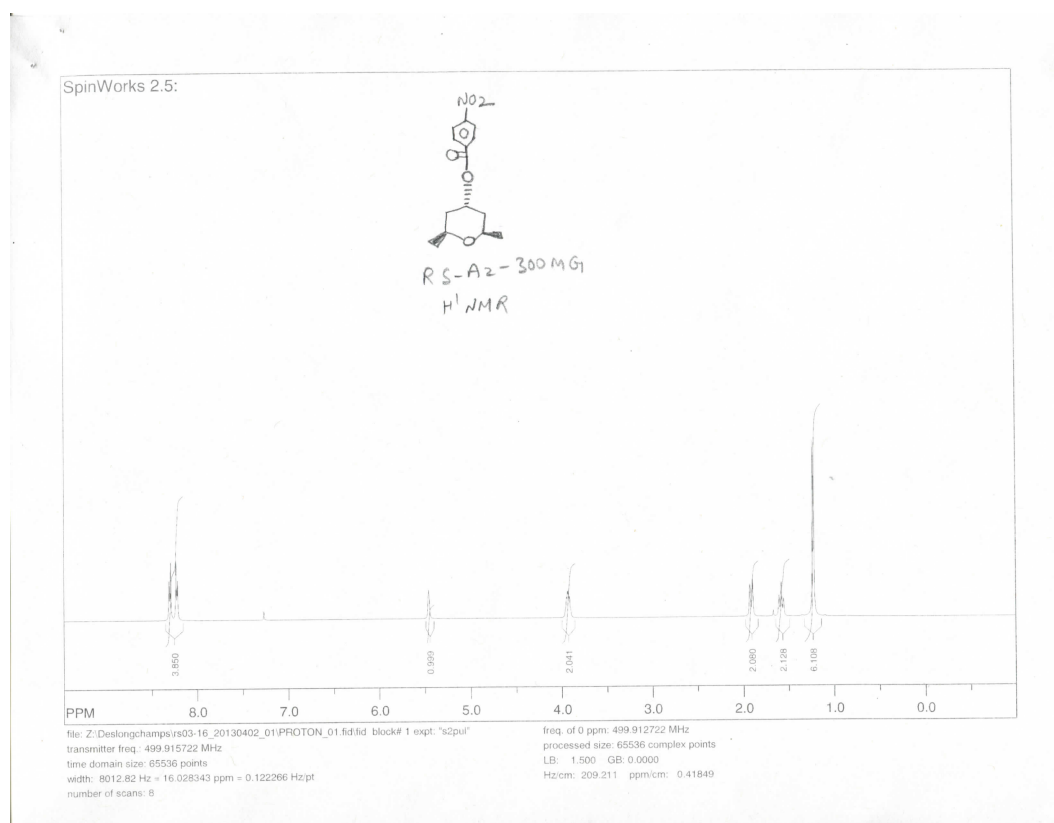
Compound 4c



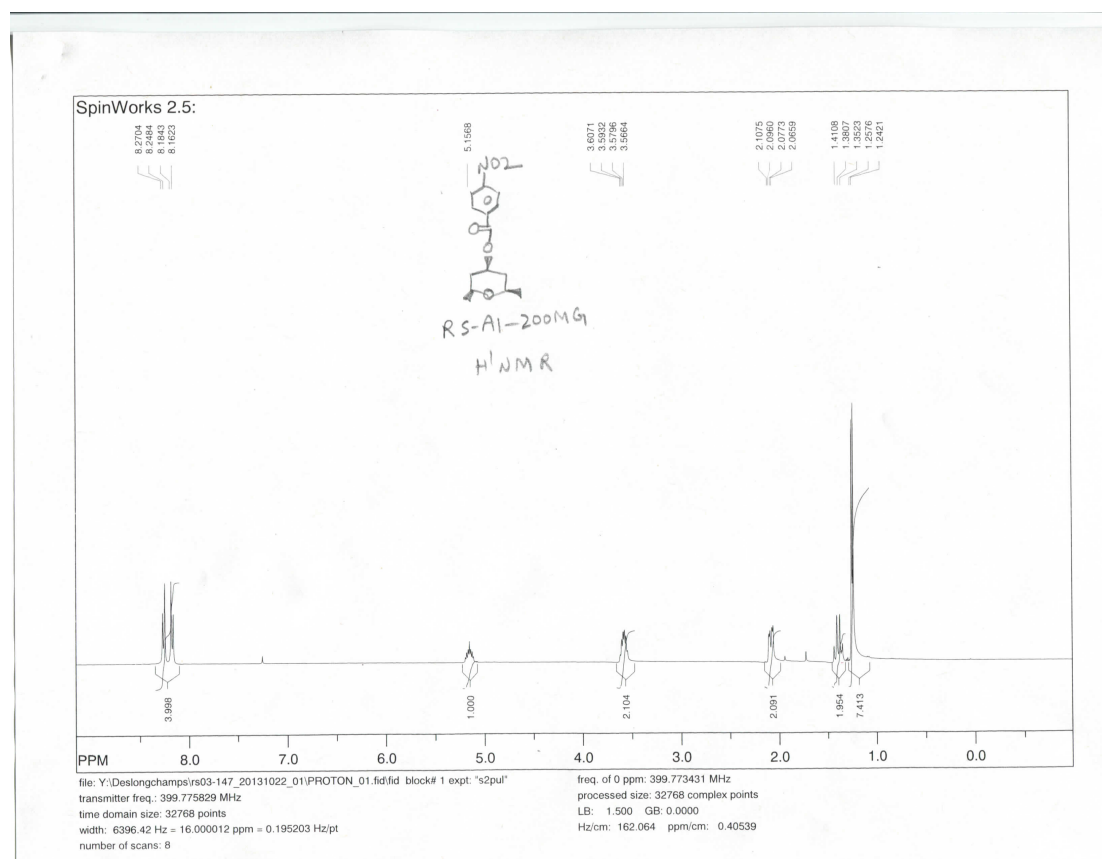
Compound 5c

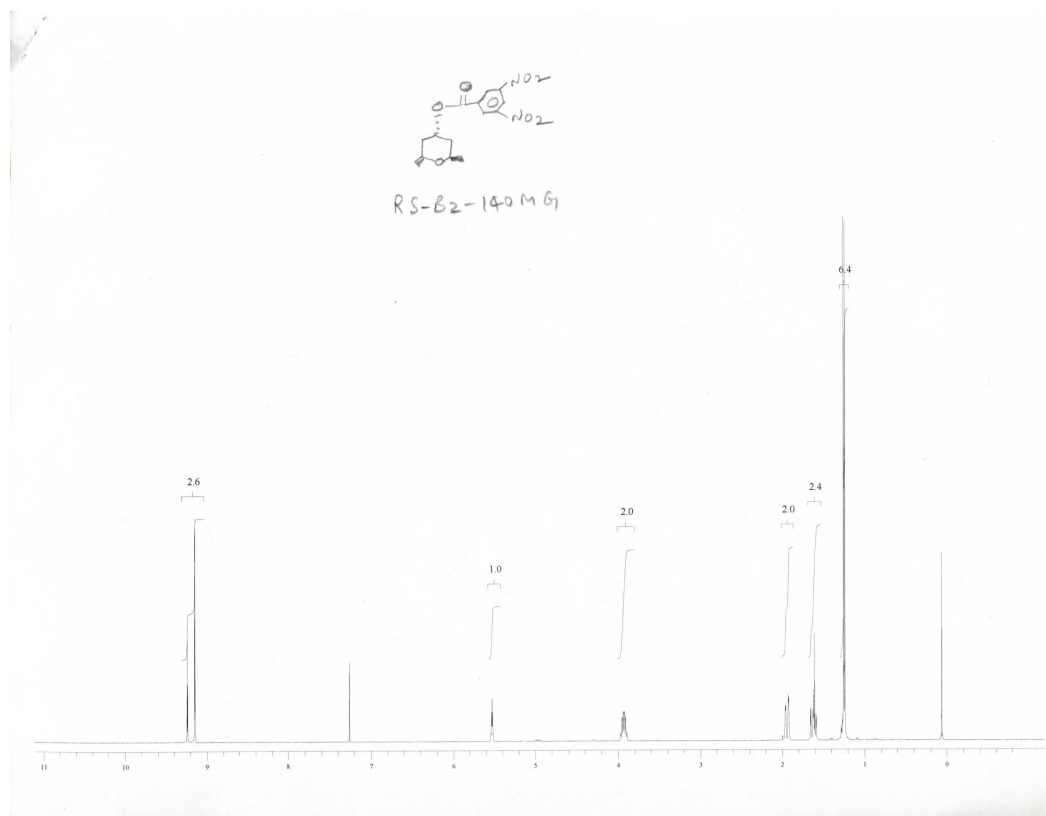
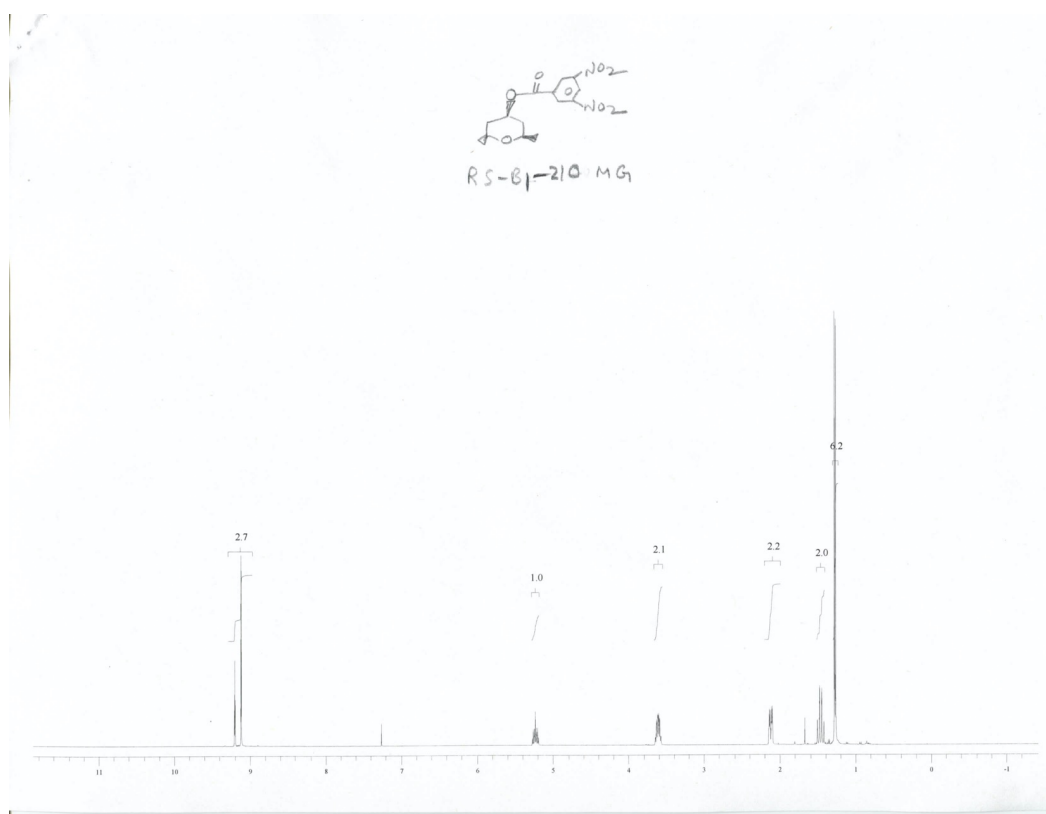


Compound 4d

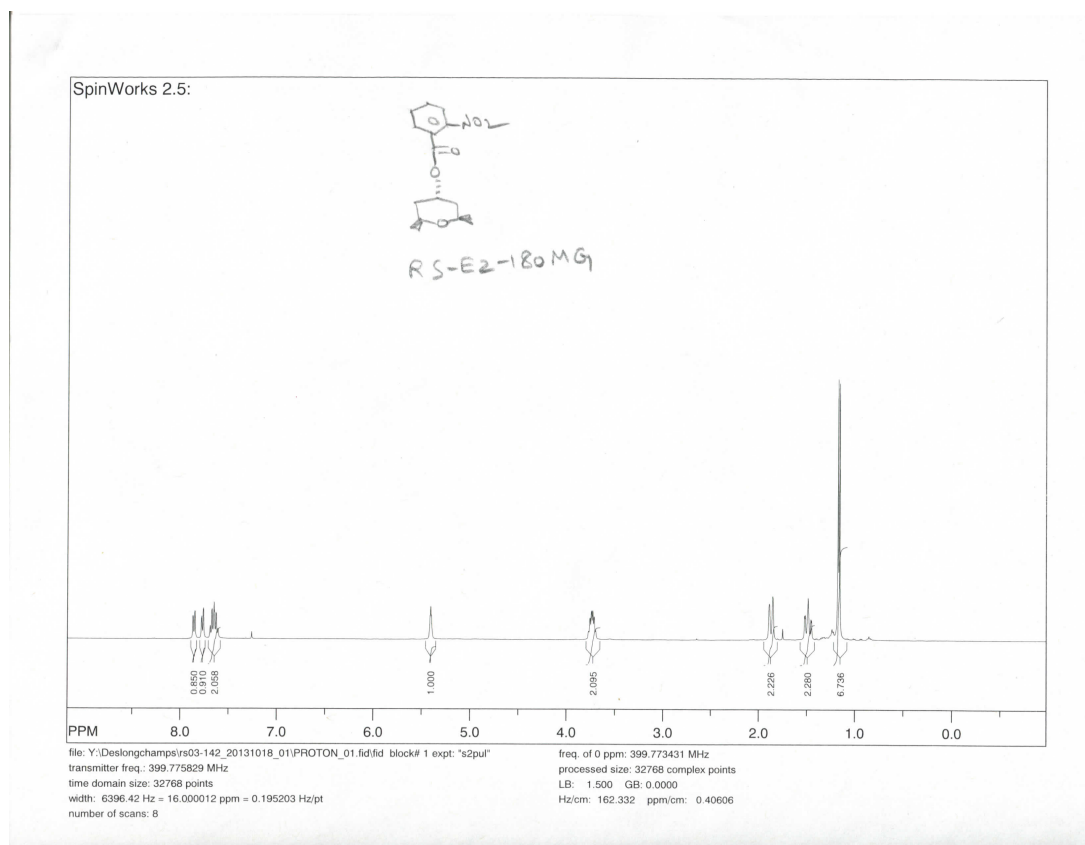


Compound 5d



Compound 4e**Compound 5e**

Compound 4f

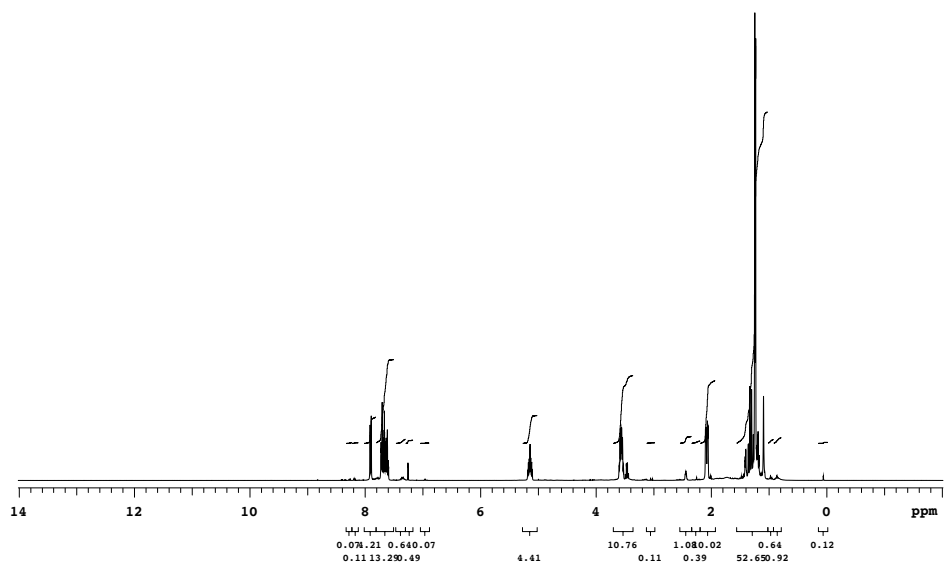


Compound 5f

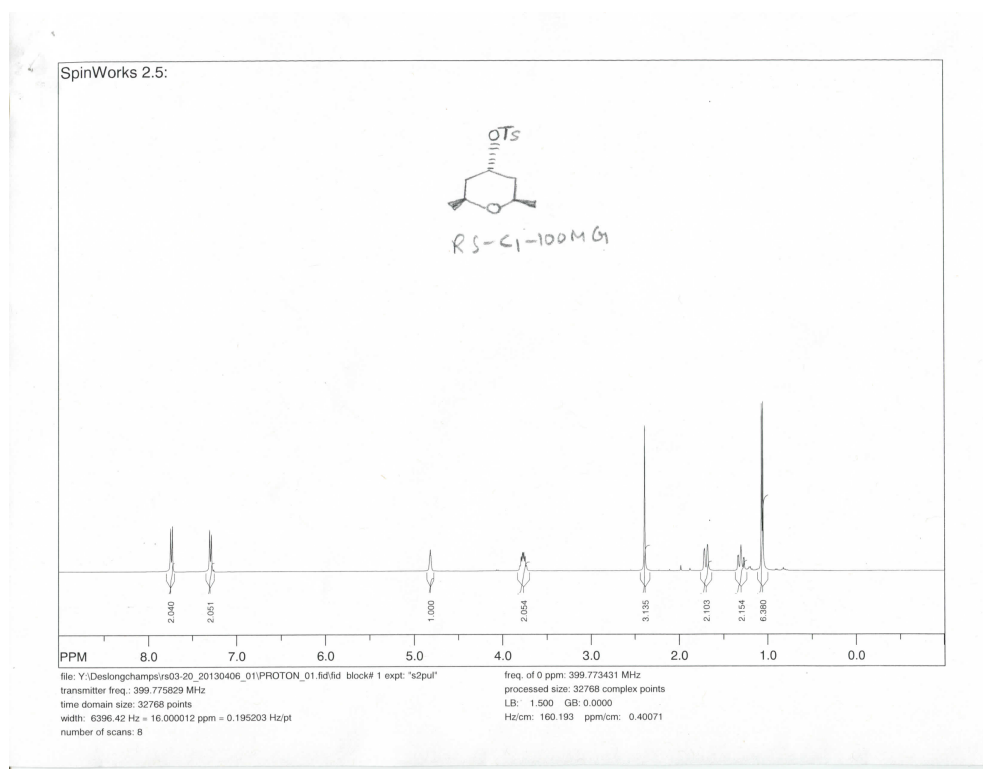


1o-1_18

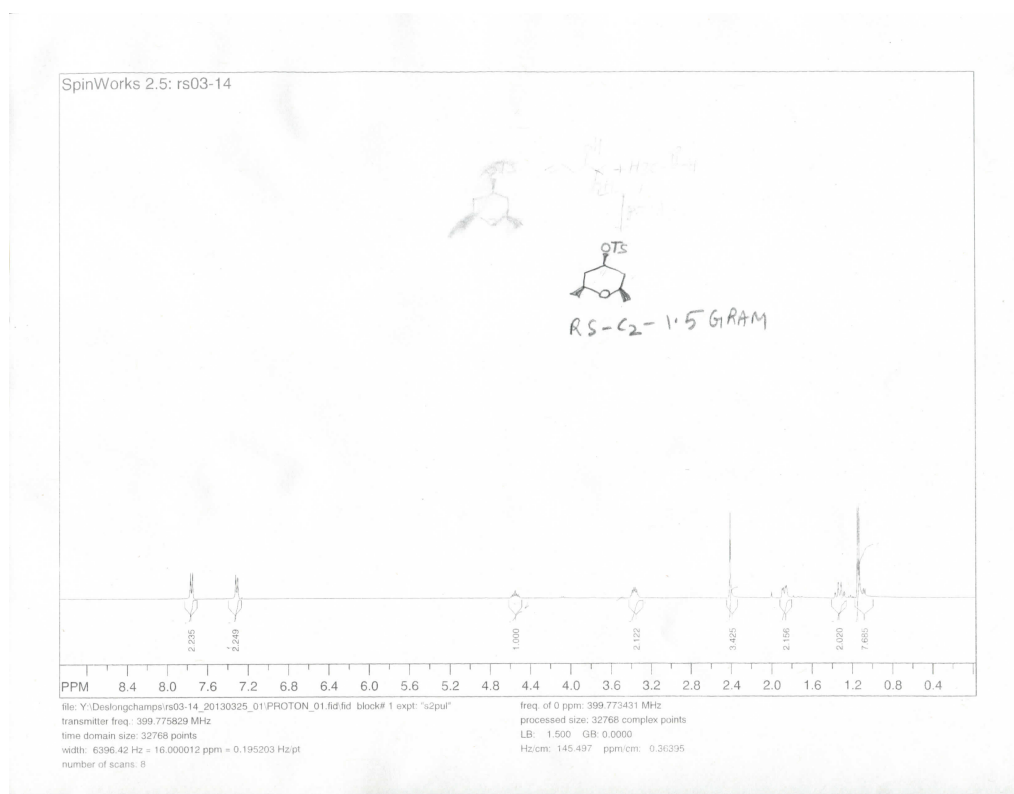
Sample Name 1o-1_18 Pulse sequence PROTON Temperature 25 Study owner jmw-lo
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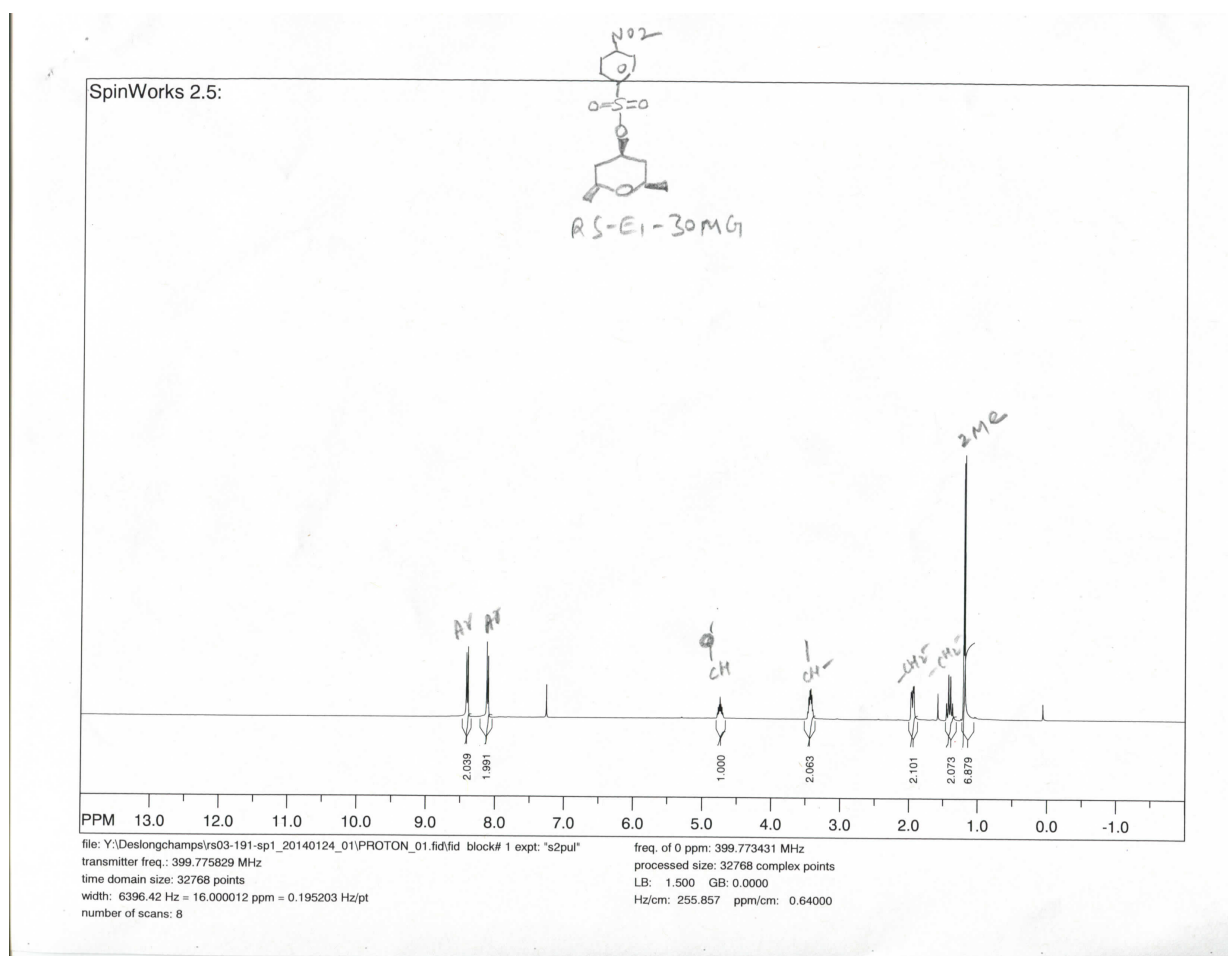
Compound 4h



Compound 5h



Compound 5i



Gaussian Cartesian Coordinates and Energies Calculated at B3LYP/6-311++G(d,p):

Energies given as sum of electronic and zero-point vibrational energies (E), sum of electronic and thermal enthalpies (H), sum of electronic and thermal Free energies (G), and zero-point vibrational energy (ZPVE) in Hartrees.

4

E = -425.546958

H = -425.536438

G = -425.580672

ZPVE = 0.205259

Symbol	X	Y	Z
O	0.03556100	2.20567600	-0.87047400
O	0.02198700	-1.25993900	-0.00407800
C	-0.02213100	1.59409600	0.42974500
H	-0.03219900	2.38116000	1.19534400
C	-1.18691600	-0.54752500	-0.28480000
H	-1.17244300	-0.24502800	-1.34266000
C	-1.26779100	0.71131400	0.58630900
H	-2.16847400	1.28524500	0.33560700
H	-1.35981200	0.41017500	1.63661700
C	1.20688300	-0.50284600	-0.28921600
H	1.17809000	-0.20111800	-1.34596900
C	1.24855900	0.75774600	0.58104000
H	1.35399400	0.45958400	1.63015200
H	2.11471300	1.37113500	0.31662500
C	-2.34116300	-1.50850700	-0.04896100
H	-2.23641100	-2.39104900	-0.68355300
H	-3.29609000	-1.02737900	-0.27853700
H	-2.35895900	-1.83663900	0.99401100
C	2.39343900	-1.42336100	-0.05540200
H	2.42687100	-1.74802300	0.98839500
H	3.32928100	-0.90873600	-0.28981500
H	2.31760000	-2.31045200	-0.68784300
H	-0.75561400	2.73945900	-0.99576000

4-H₂O⁺

E = -425.851573

H = -425.840257

G = -425.886039

ZPVE = 0.216107

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C	-0.00725000	1.48682200	0.63234800
H	-0.01463000	2.33311800	1.31807200
C	-1.18916800	-0.55922800	-0.29456700
H	-1.16105700	-0.23030200	-1.34482700
C	-1.27530300	0.67726500	0.62151600
H	-2.14922400	1.29924000	0.39817900
H	-1.40384200	0.34140200	1.65921300
C	1.20508300	-0.54505100	-0.28641900
H	1.18031200	-0.21645600	-1.33687100
C	1.27040800	0.69239100	0.63018100
H	1.39566600	0.35799100	1.66873900
H	2.13839300	1.32475900	0.41292800
C	-2.35847600	-1.50637200	-0.09173500
H	-2.25882400	-2.36551400	-0.75586000
H	-3.30576700	-1.00812500	-0.31289900
H	-2.38459700	-1.86978400	0.93826200
C	2.38424200	-1.47814000	-0.07550700
H	2.40766600	-1.84114000	0.95469900
H	3.32698000	-0.96857800	-0.29026800
H	2.29942400	-2.33845600	-0.74017200
H	0.78502500	2.82884300	-0.97508300
H	-0.80291100	2.82175700	-0.97971600

5

E = -425.547872

H = -425.537276

G = -425.581692

ZPVE = 0.205119

Symbol	X	Y	Z
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O	-2.87026400	-0.00557700	-0.11326800
C	0.71099900	1.19945000	0.33252000
H	0.74306000	1.19036200	1.43516500
C	-0.75673400	1.25248800	-0.11297000
H	-1.24550400	2.14224100	0.29538600
H	-0.78646000	1.32875100	-1.20805400
C	-1.51080400	-0.00294100	0.32951500
H	-1.58938000	-0.00319900	1.42200300
C	0.71568200	-1.19664100	0.33232600
H	0.74768900	-1.18758900	1.43497400
C	-0.75182500	-1.25533300	-0.11321600
H	-0.78124600	-1.33150400	-1.20830700
H	-1.23714100	-2.14705500	0.29495800
C	1.53467200	2.37019800	-0.17968400
H	2.56866400	2.28678000	0.16132600
H	1.12347200	3.31585400	0.18331700
H	1.53396300	2.38701400	-1.27277600
C	1.54395000	-2.36407300	-0.18002400
H	1.54318400	-2.38087700	-1.27311600
H	1.13656600	-3.31136900	0.18300200
H	2.57763900	-2.27653700	0.16087200
H	-2.87699300	-0.00437900	-1.07704900

5-H₂O⁺

E = -425.854075

H = -425.842602

G = -425.888806

ZPVE = 0.216351

Symbol	X	Y	Z
O	-1.36930800	0.01969500	-0.17725900
O	2.92916800	-0.10150500	-0.15621200
C	-0.81321100	-1.17866600	0.34075800
H	-0.86752200	-1.15901700	1.43920600
C	0.69172800	-1.26924200	-0.08069000
H	1.14940600	-2.15892900	0.35515400
C	1.35241600	-0.01474800	0.39521600
H	1.58400000	-0.02258600	1.45938100
C	-0.78327300	1.20325600	0.33717800
H	-0.83552600	1.18812400	1.43592300
C	0.72439900	1.25397600	-0.08919800
H	0.75258500	1.32061100	-1.18077100
H	1.19439200	2.14037300	0.34439900
C	-1.61440400	-2.35319700	-0.19238800
H	-2.65373800	-2.26020800	0.12490900
H	-1.21985200	-3.29626800	0.19198200
H	-1.58790500	-2.37518500	-1.28399700
C	-1.55407200	2.39744600	-0.19778800
H	-1.52730600	2.41679900	-1.28940600
H	-1.13637600	3.33119100	0.18511100
H	-2.59535900	2.33100800	0.11976400
H	2.98280500	-0.31003600	-1.10558700
H	0.71517500	-1.34077700	-1.17329200
H	3.45458300	0.69955500	0.01946600

6

E = -389.609996

H = -389.599237

G = -389.643880

ZPVE = 0.229258

Symbol	X	Y	Z
O	-0.04660100	2.18113200	-0.79413400
C	0.01290400	1.54933200	0.49751500
H	0.01299800	2.32857400	1.27244400
C	-1.27429000	-0.51513800	-0.28956800
H	-1.27134000	-0.14557400	-1.32186900
C	-1.25674500	0.71001100	0.63968200
H	-2.12439800	1.34823900	0.44566500
H	-1.33132200	0.37525500	1.68268100
C	1.28684100	-0.53508000	-0.27857800
H	1.29533000	-0.16890500	-1.31270700
C	1.27550900	0.69064900	0.64968800
H	1.34054300	0.35690400	1.69432900
H	2.16431700	1.31022100	0.47150600
C	-2.54191000	-1.35023000	-0.08367000
H	-2.56893200	-2.20445500	-0.76712500
H	-3.44293700	-0.75445800	-0.25880700
H	-2.59478900	-1.74103000	0.93883600
C	2.54133800	-1.38767100	-0.06333500
H	2.57935000	-1.78116000	0.95859100
H	3.45332100	-0.80542600	-0.22897500
H	2.56203400	-2.24085800	-0.74814100
H	0.74148800	2.72213500	-0.90701800
C	-0.00097300	-1.35307400	-0.08650200
H	-0.00341000	-2.20384900	-0.77818300
H	-0.00791800	-1.77947200	0.92819400

6-H₂O⁺

E = -389.921079

H = -389.909514

G = -389.955751

ZPVE = 0.240467

Symbol	X	Y	Z
O	0.01740500	2.24446000	-0.85494100
C	0.00764000	1.47899500	0.62209000
H	0.00903900	2.34090500	1.28897400
C	-1.28172300	-0.53434900	-0.29611000
H	-1.28278600	-0.17697100	-1.33414400
C	-1.27640500	0.69578100	0.63207300
H	-2.13658800	1.35213600	0.45394900
H	-1.38293900	0.36961400	1.67738400
C	1.28513400	-0.54671700	-0.28550600
H	1.29822100	-0.18943600	-1.32349100
C	1.28397400	0.68346300	0.64262200
H	1.37864800	0.35637400	1.68878400
H	2.15188800	1.33150000	0.47166400
C	-2.55253500	-1.36199000	-0.07993000
H	-2.57996400	-2.21113500	-0.76598300
H	-3.45466700	-0.76841700	-0.25179000
H	-2.59814600	-1.75617800	0.94000200
C	2.54610600	-1.38650400	-0.05877900
H	2.57945000	-1.78101000	0.96150400
H	3.45529400	-0.80162900	-0.22312800
H	2.57110700	-2.23593600	-0.74456700
H	0.81486800	2.77925400	-1.01106500
H	-0.77295200	2.78786600	-1.01732400
C	-0.00310300	-1.36137100	-0.08292400
H	-0.00433400	-2.21127400	-0.77094300
H	-0.00932100	-1.78661800	0.93016300

7

E = -389.610813

H = -389.599988

G = -389.644788

ZPVE = 0.229204

Symbol	X	Y	Z
O	-0.00064600	2.86140400	0.08450400
C	-1.28224000	-0.71237800	-0.33471200
H	-1.29483100	-0.71565300	-1.43420400
C	-1.26326100	0.75579000	0.12511200
H	-2.14425700	1.28827600	-0.24714600
C	-0.00033700	1.49083800	-0.33113700
H	-0.00034400	1.54900400	-1.42499100
C	1.28255500	-0.71179800	-0.33470700
H	1.29514300	-0.71506200	-1.43420000
C	1.26291200	0.75635700	0.12513000
H	1.31273200	0.78822900	1.22420900
H	2.14367000	1.28924700	-0.24711300
C	-2.54180900	-1.43798100	0.14997900
H	-2.56787600	-2.47093400	-0.20987900
H	-3.44893500	-0.93793900	-0.20222200
H	-2.58052500	-1.46590900	1.24436800
C	2.54245000	-1.43683800	0.14997500
H	2.58118300	-1.46475300	1.24436400
H	3.44934900	-0.93638500	-0.20222400
H	2.56898400	-2.46977800	-0.20988600
H	-0.00098000	2.88666200	1.04792300
H	-1.31310500	0.78764100	1.22419300
C	0.00031800	-1.43048900	0.11772800
H	0.00055300	-2.45939300	-0.26049900
H	0.00033100	-1.50412500	1.21511100

7-H₂O⁺

E = -389.922171

H = -389.910533

G = -389.956983

ZPVE = 0.240882

Symbol	X	Y	Z
O	-2.81414500	-0.74147400	0.11460100
C	1.06409400	-1.06019600	-0.33994700
H	1.08448700	-1.07274200	-1.43731500
C	-0.39046900	-1.40067900	0.10364700
H	-0.68976400	-2.37228800	-0.29548500
C	-1.29360800	-0.31052800	-0.39064200
H	-1.48372700	-0.36104300	-1.46195700
C	0.45983600	1.43391500	-0.33689500
H	0.46987800	1.45491500	-1.43431600
C	-0.98791600	1.06920200	0.11269100
H	-1.03121800	1.09567400	1.20776300
H	-1.69197500	1.80745200	-0.28316200
C	2.03323000	-2.13274800	0.16683300
H	3.04764000	-1.92077200	-0.17771200
H	1.75945100	-3.12555900	-0.19835500
H	2.05153700	-2.16455500	1.26035200
C	0.83204300	2.83060000	0.17048700
H	0.83490600	2.86633400	1.26397700
H	0.13499600	3.58950700	-0.19348500
H	1.83088300	3.10566600	-0.17485500
H	-2.83453400	-1.00653000	1.05090000
H	-0.41292700	-1.44709400	1.20022900
H	-3.48590300	-0.05216900	-0.03248500
C	1.44364900	0.35136700	0.13314200
H	2.44456600	0.59483300	-0.23398400
H	1.50563900	0.36534700	1.22956700

References

¹ C.S. Barry, N. Bushby, J.R. Harding, R.A. Hughes, G.D. Parker, R. Loe, C.L. Willis, Chem. Comm. **2005**, 3727-3729

² J.A. Bender, O.D. Lopez, G. Wang, M. Belema, J.F. Kadow, WO 2012/109080 A1