



A comparative study of the combustion dynamics and flame properties of dead Mediterranean plants

A. Sahila^A , H. Boutchiche^A, D. X. Viegas^B, L. Reis^B and N. Zekri^{A,*} 

For full list of author affiliations and declarations see end of paper

***Correspondence to:**

N. Zekri

Faculty of Physics, LEPM (Laboratoire d'Etude Physique des Matériaux), Université des Sciences et de la Technologie d'Oran, LEPM BP 1505 El Mnaouer, 31000, Oran, Algeria
Email: nouredine.zekri@univ-usto.dz

ABSTRACT

Background. The physical processes governing flame behaviour are key elements for a better understanding of forest fires. **Aims.** To study the combustion properties of several dead Mediterranean forest fuels. **Method.** Samples of straw, eucalyptus, shrubs and *Pinus Pinaster* with the same load were placed in circular containers of the same size, and ignited in the absence of wind. **Key results.** The combustion parameters (burning rate, flame height, temperature and gas velocity) evolved according to the same trend regardless of the fuel type. A new law is proposed to account for the anomalous relaxation processes occurring in the growth and decay phases of the flame. The dynamic exponent depends on the vegetation type only in the growth phase (highest for *Pinus Pinaster* and lowest for straw). The relaxation times are shortest for shrubs and largest for straw. The maximum flame height and burning rate are largest for shrubs and lowest for straw. Froude modelling suggests that the scaling behaviour of the flame may depend on the fuel type. **Conclusions.** The observed relaxation parameters driving fire dynamics and the combustion characteristics depend on the nature of the fuel. **Implications.** Further investigation of the vegetation region's influence on these properties is necessary.

Keywords: air velocity and temperature profiles, anomalous diffusion, anomalous relaxation, flame height, flaming combustion, forest fires, heat release rate, Mediterranean plants, turbulent diffusion flame.

Introduction

The increasing intensity and severity of forest fires around the globe represent a major concern because they seriously endanger the ecosystem by affecting flora and fauna and the environment (Pausas *et al.* 2008; Vilén and Fernandes 2011). Therefore, a deeper understanding of the combustion dynamics of forest fuels and a better estimation of their combustion characteristics and their correlations are necessary. Forest fire is a very complex phenomenon involving physical processes that occur at different spatial scales ranging from the microscopic (smallest) scale, where the three phases of the fuel can be distinguished (solid, liquid and gas), to the gigascopic (largest) scale, where a fractal analysis of its pattern becomes feasible (Séro-Guillaume and Margerit 2002; Sahila *et al.* 2021). At intermediate scales, many works have already been devoted to examining the behaviour of these parameters in the case of pool fires (Zabetakis and Burgess 1961; Tarifa 1967; Kung and Stavrianidis 1982; Babrauskas 1983; Koseki and Yumoto 1988; Koseki 1989; Klassen and Gore 1994; Chatris *et al.* 2001), fire whirls (Martin *et al.* 1976; Lei *et al.* 2011; Pinto *et al.* 2017) and natural fires (Thomas 1963; Dupuy *et al.* 2003; Sun *et al.* 2006; Sahila *et al.* 2023). In the present work, an experimental study of the burning characteristics of several Mediterranean forest fuels was carried out where the turbulent diffusion flame was subjected only to buoyancy forces. Samples of dead shrubs (mix of heather (*Erica australis*) and gorse (*Pterospartum tridentatum*)), *Pinus Pinaster* needles, *Eucalyptus globulus* leaves and *Avena sativa* straw were placed in cylindrical baskets of the same size and ignited to study the fuel type's effect on fire behaviour. These containers can represent single fuel bed items (e.g. trees) in fire spread modelling. For

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instance, using the so-called small-world-network model, the fuel bed items were considered cylindrical to simulate wildfire spread (Adou et al. 2010). However, in this model, as well as in many other semi-physical models (e.g. Rothermel's model (1972)), the heat released by the burning items is assumed to be independent of time, which does not account for the different burning phases of these items (growth, fully developed and decay; see Manzello 2020). Only computational fluid dynamics (CFD) models estimate the time-dependent burning rate by using a set of physical equations. Indeed, the dynamics of a fluid flow can be described by a set of differential equations expressing the conservation of mass, momentum, energy and chemical species (see table 1 of Williams (1969)). When considering a multi-component fluid with various chemical species that interact, the mass conservation of a particular species i ($i \in F$, F being the set of all the chemical species constituting the fluid) is given by the advection–diffusion equation (de Groot and Mazur 1984; Rosner 1986; Kee et al. 2003; Manzello 2020):

$$\frac{\partial c_i}{\partial t} + \nabla \times (c_i \mathbf{v}) = \nabla(D \nabla c_i) + \omega_i \quad (1)$$

where $c_i = \rho_i x_i$ is the concentration of species i and x_i its mass fraction, D is the diffusion coefficient and ω_i the production rate of the species by chemical reactions. By summing all the species' mass conservation equations, we recover the fluid's mass conservation equation (the well-known continuity equation), where the sum of all the mass production and destruction rates and diffusion fluxes is null (Kuwana 2019). From a statistical physics point of view, this advection–diffusion equation corresponds to the Fokker–Planck equation, describing the time evolution of the particle velocity probability density function under the influence of drag forces, which can be deduced from a generalised/anisotropic random walk model. The diffusion term in the right side of Eqn 1 is provided by Fick's law, which is valid only when the gas particles exhibit normal (Brownian) diffusion. In this case, their Mean Square Displacement (MSD) is given by $\langle \Delta x^2 \rangle \propto Dt$, where $D = RT/6\pi\eta r N_A$ with R , T , η , r and N_A being the universal gas constant (8.314 J/mol*K), temperature (in Kelvin), dynamic viscosity, radius of the particle and Avogadro's number respectively. A normal diffusion was found to be related to an exponential relaxation (Metzler and Klafter 2000; Bouchaud 2008). Relaxation means that the system moves irreversibly towards an equilibrium state under the application of an external force (Kremer and Schonhals 2003). In the case of forest fires, it is induced by the existence of storing (heating of the fuels) and dissipative (heat release/particle emission from the burning vegetation) processes. The relaxation process obeys the following differential equation (Stanislavski and Weron 2010):

$$\frac{\partial \phi}{\partial t} = a(t)\phi(t) \quad (2)$$

where $\phi(t)$ is the relaxation function with the boundary conditions $\phi(t = 0) = 1$ and $\phi(t = \infty) = 0$. $\phi(t)$ is usually given, in its general form, by the empirical Kohlrausch–Williams–Watts equation (Kohlrausch 1854; Williams and Watts 1970):

$$\phi(t) = e^{-\left(\frac{t}{\tau}\right)^\beta} \quad (3)$$

Here, τ is the relaxation time. The exponential relaxation, associated with Brownian diffusion, occurs only when $a(t) = 1/\tau$ is constant in Eqn 2, corresponding to $\beta = 1$, where β is the stretching exponent. In most cases, heterogeneous complex systems exhibit a non-exponential relaxation process ($\beta \neq 1$), corresponding to a broad distribution of exponential-like relaxation times (Bouchaud and Georges 1990; Metzler and Klafter 2000; Morgado et al. 2002; Bouchaud 2008). If $\beta < 1$, the system exhibits a stretched exponential relaxation that is associated with a 'sub-diffusion' process (Metzler and Klafter 2000). In the case of forest fire spread, sub-diffusion may correspond to an anomalously slow propagation of fire induced by the existence of a broad distribution of ignition times (trapping times) (Sahila et al. 2021). If $\beta > 1$, there is a compressed exponential relaxation related to a 'super-diffusion' process (Bouchaud 2008) that may be induced by the existence of long-range physical effects beyond the primary combustion zone during fire spread (Porterie et al. 2008). These anomalous diffusion processes do not follow Fick's law, and the linear relationship between the MSD of the particles and time occurring for normal diffusion (Oliveira et al. 2019) can be affected significantly by disorder in a real medium (e.g. porous fuel), which renders the temporal evolution of these particles' MSD non-linear, $\Delta x^2 \sim t^\alpha$, with α a real positive number (Metzler and Klafter 2000; Morgado et al. 2002) discriminating between sub-diffusion ($\alpha < 1$), Brownian/normal diffusion ($\alpha = 1$) and superdiffusion ($\alpha > 1$) regimes. However, the first four equations listed in table 1 of Williams (1969) used by CFD models consider only normal diffusion processes, and do not account for the trapping effect of flammable volatiles by fuel particles (in the condensed phase) and buoyant gases (in the gas phase) or their jump caused by the coherent flow leading to anomalous diffusion. Anomalous diffusion and relaxation processes have been observed in various systems (Richardson 1926; Scher and Montroll 1975; Koch and Brady 1988; Bunde and Havlin 1996; Küntz and Lavallée 2001; Bennett et al. 2003; Dobrovolskis et al. 2007; Matthäus et al. 2009; Llievski et al. 2018). Sahila et al. (2023) found recently that these anomalous processes drive the burning dynamics of dead straw, and they introduced a Kohlrausch–Williams–Watts (KWW) equation to accurately describe the time evolution of the source's heat flux.

These processes are examined in the present work for the above-mentioned Mediterranean plants to characterise their burning dynamics through their relaxation properties, and to check the universality of these dynamics for their use in

spread modelling. Indeed, it is well known that pyrolysis gas particles move from a region of higher concentration to a region of lower one, but what is the nature of this diffusion, and is it dependent on the vegetation type? Do they diffuse in the same way during the growth and decay phases of the flame? These questions are of fundamental importance for an accurate understanding of turbulent diffusion flame dynamics and will be addressed in this work. Moreover, the time evolution of the mass-loss rate, flame height and temperature, and upward gas velocity are examined during all the phases of fire development. A systematic comparison between the combustion characteristics and relaxation properties of these fuels is carried out to study their correlations and their role in flaming combustion.

Experimental setup

Samples of dead shrubs (mix of heather (*Erica australis*) and gorse (*Pterospartum tridentatum*)), *Pinus Pinaster* needles, *Eucalyptus globulus* leaves and *Avena sativa* straw with the same initial fuel mass (M_f 1.56 kg), height (0.34 m) and bulk density (ρ_f 23.4 kg/m³) were placed in cylindrical containers (with a fixed diameter d_c 0.5 m) made of a metallic grid and open at the top. Before each burning experiment, the fuel moisture content was estimated using a moisture analyser A&D MX-50 with a resolution of 0.01%. The relative humidity of the air and ambient temperature were measured by a thermo-hygrometer. The flame is subdivided into three regions (McCaffrey 1979): (i) a continuous zone that begins at the fuel surface and where the velocity v of gas particles increases with height z ($v \propto \sqrt{z}$) and the temperature is constant ($T \propto z^0$); (ii) the intermittent region (the pulsating part of the flame) where v is approximately invariant and the flame temperature decreases inversely with height ($T \propto z^{-1}$); (iii) the thermal plume (situated above the flame tip) where the gas velocity begins to decrease with height ($v \propto z^{-1/3}$) and the smoke temperature continues falling at a faster rate ($T \propto z^{-5/3}$). A K-type thermocouple (nickel–chromium/nickel aluminium, metallic shielded, with a diameter of 0.5 mm) and a S-Pitot tube were placed 1 m above the fuel surface to determine the flame temperature and air velocity at this height respectively (see Fig. 1). It is difficult to perform corrections for the thermocouple measurements owing to radiation losses because for certain periods of time, the thermocouple was surrounded by the flames – in which case there are no radiative losses – whereas in other periods, it was intermittently inside the flames, and in others it was placed at variable distances from the flame. Owing to these errors, we assume that the absolute values of the measured temperatures are not accurate. Nevertheless, the relative temporal variation of the temperature indicates a pattern that is consistent with the expected trend (higher values for lower relative positions and vice versa).

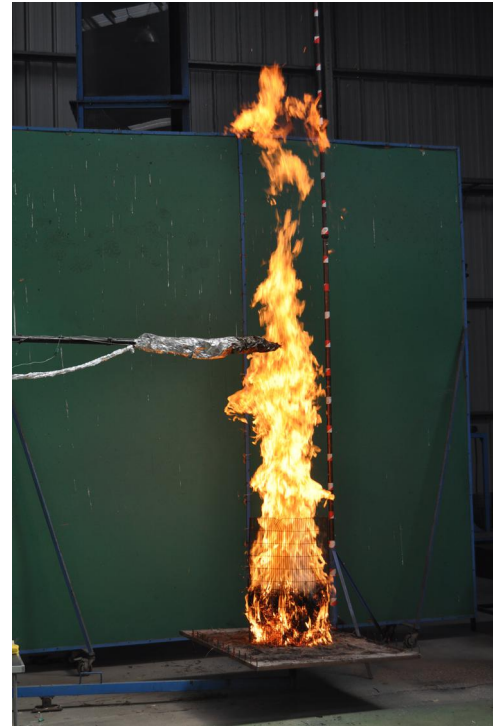


Fig. 1. Experimental setup for straw burning experiment at the ADAI fire laboratory (Coimbra).

At the beginning of each test, the lowest region of the basket was ignited by a gas burner, and three experiments were carried out for each fuel to ensure the repeatability of the tests. Note that each test corresponds to a different configuration of the porous fuel bed, which induces statistical fluctuations in the measured parameters (mass-loss rate and flame height). The three replicates allowed relative fluctuations smaller than 20% to be obtained (see the Results section below), which are estimated as the tolerance of the measurement due to statistical errors (not only to the limited precision of the apparatus). Each experiment was recorded by an optical video camera (Sony FDR-AX53) and a digital camera (Canon EOS 550D). The videos were segmented into images using video to JPG converter software, allowing the flame heights during the entire duration of flaming combustion to be estimated. The flame height was measured manually as the distance from the flame bottom to its visual tip. Hence, a vertical scale was placed near the container to allow proper scaling of the flame height values (the rubbers are equidistant by 0.2 m). A digital balance A&D HW-100KGL (10 g resolution) with a frequency of 1 Hz was used to measure the mass loss of the fuels during their combustion, and values were recorded on computer by RSKey v.1.40 software.

Results and discussion

The time evolution of the normalised fuel mass (M_f/M_0 , M_0 being the initial mass of the fuel) is shown in Fig. 2a for all

the fuels considered. The fuel mass decreases towards an asymptotic minimal value M_{fmin} , and this tendency to equilibrium seems to depend on the fuel type: it is faster for shrubs than for *Pinus Pinaster* needles (PP), straw and Eucalyptus leaves (Euc). These curves can be described by the following equation (Drysdale 2011):

$$\frac{dM_f}{dt} = -K(T) \times M_f \quad (4)$$

where $K(T)$ obeys the Arrhenius law $K(T) \propto \exp(-E_A/RT)$. E_A is the activation energy (J/mol) and R the universal gas constant (8.314 J/K mol). This formula is very simple because it takes into consideration only water evaporation accompanied by volatiles emission during the combustion of the fuel. It does not take into consideration other mechanisms such as the decomposition and formation of molecules (David 1975). In earlier work, Sahila et al. (2023) studied the combustion characteristics of dead straw for different container sizes, and demonstrated that Eqn 4 can also describe a relaxation process where M_f is replaced by a relaxation function $\phi(t) = (M_f(t) - M_{fmin})/(M_0(t) - M_{fmin})$ obeying Eqn 2 with $K = 1/\tau$, τ being a relaxation time that measures the rate of decrease of the relaxation function (see Eqn 3), i.e. the rate of consumption of the fuel during a characteristic period (growth or decay). If K (hence, the temperature) is constant, $\phi(t)$ decreases exponentially over time, describing normal relaxation ($\beta = 1$). However, as temperature varies with time, one expects that heterogeneous porous fuels exhibit non-exponential relaxation processes described by Eqn 3. To verify the existence of such anomalous dynamic physical processes for all the Mediterranean vegetation species studied, it is more suitable to use $-\ln(\phi(t))$:

$$\ln\left(\frac{M_0 - M_{fmin}}{M_f(t) - M_{fmin}}\right) = \left(\frac{t}{\tau}\right)^\beta \quad (5)$$

Hence, by plotting $-\ln(\phi(t))$ in logarithmic scales, the dynamic exponent β and the relaxation time τ can be deduced directly from the linear fit, as shown in Fig. 2b for dead shrubs.

For short times, the burning fuel's mass exhibits super-relaxation through time (the first slope β_1 is greater than unity), characterising the growth phase of the turbulent diffusion flame. Then, a transition zone characterising a crossover time t_x appears, corresponding to the fully developed phase of the flame. Finally, during the decay phase of the flame (long times), the relaxation process becomes anomalously slow (the second slope β_2 is smaller than unity). The curves for all the vegetation types studied exhibit the same trend.

The dynamic exponents (β_1 and β_2) and the relaxation times (τ_1 and τ_2) obtained for each vegetation type in the two regimes are summarised in Table 1. The results show that the growth phase is characterised by a super-relaxation process ($\beta_1 > 1$) of the burning fuel's mass, while the decay phase is characterised by an anomalously slow relaxation process ($\beta_1 < 1$) regardless of the vegetation type for all experiments. Hence, it seems that these anomalous processes are universal, and inherently characterise the various phases of flaming combustion. β_1 varies from one vegetation type to another; it is highest for *Pinus Pinaster* and lowest for straw. However, β_2 seems to be independent of the vegetation type. The relaxation times (τ_1 and τ_2) are shorter for shrubs than for *Pinus Pinaster* needles, eucalyptus leaves and straw.

The mass-loss rate M_f is a measure of the rate at which the fuel is consumed, and is a key element to understanding the combustion dynamics. It was deduced from the balance data and its averaged values were estimated at regular intervals (5 s). The results illustrated in Fig. 3 show the same temporal trend as for the flame height, mass loss rate, temperature at thermocouple position and gas velocity

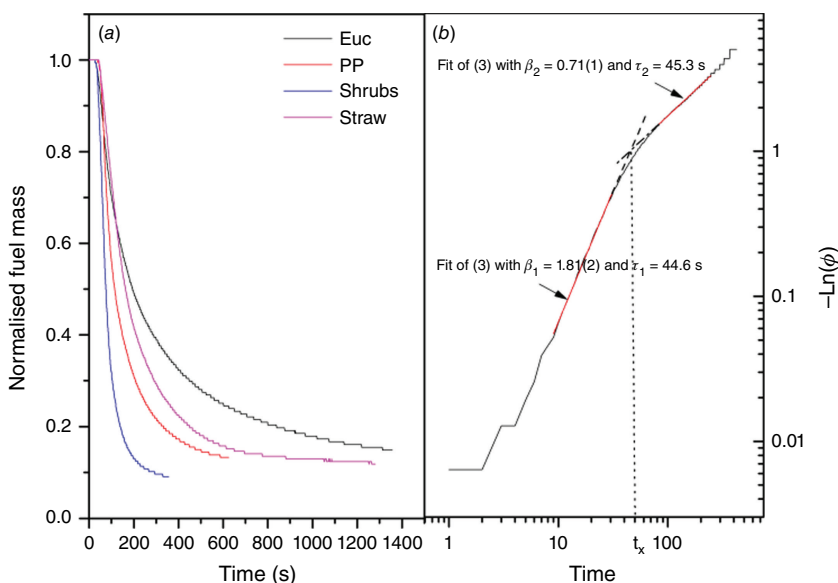


Fig. 2. (a) Temporal evolution of the normalised fuel mass for several dead forest fuels; (b) $-\ln\phi$ as a function of time in a double logarithmic scale (shrubs).

Table 1. Fitting parameters of Eqn 3 for different vegetation types.

Fuel	Test	β_1	τ_1	R_1^2	β_2	τ_2	R_2^2
Shrubs	N01	1.81 ± 0.02	44.6	0.99	0.71 ± 0.01	45.3	0.99
	N02	1.88 ± 0.01	53.2	0.99	0.93 ± 0.01	50.5	0.99
	N03	1.67 ± 0.01	41.7	0.99	0.78 ± 0.01	37.9	0.99
<i>Pinus Pinaster</i>	N01	2.34 ± 0.02	57.7	0.99	0.85 ± 0.01	76.8	0.99
	N02	2.1 ± 0.03	55.2	0.99	0.79 ± 0.01	89.8	0.99
	N03	2.35 ± 0.03	61.3	0.99	0.87 ± 0.01	93.6	0.99
<i>Eucalyptus</i>	N01	2.24 ± 0.03	90.7	0.99	0.74 ± 0.01	146.9	0.99
	N02	1.59 ± 0.02	109.9	0.99	0.73 ± 0.01	200.1	0.99
	N03	1.75 ± 0.02	90.8	0.99	0.82 ± 0.01	129.7	0.99
Straw	N01	1.18 ± 0.01	136.1	0.99	0.83 ± 0.01	144.2	0.99
	N02	1.33 ± 0.01	131.4	0.99	0.8 ± 0.01	147.8	0.99
	N03	1.37 ± 0.01	183.9	0.99	0.88 ± 0.01	201.6	0.99

at Pitot tube position. Fire development is as follows: after ignition, the growth phase begins, where the flame spreads vertically very rapidly along the lateral fuel bed surface; then, a consistent flame is formed, accompanied by horizontal spread of the flame through the upper surface of the fuel. In this phase, the flame height, mass loss rate, temperature and upward gas velocity increase continuously until they attain their peak. The above observed super-relaxation (see Fig. 2b and Table 1) corresponds to super-diffusion of the emitted gas particles ($\alpha > 1$). Gas particles are mainly emitted from the fuel surface in the growth phase, and the absence of obstacles to their motion in addition to the fact that they are coherently convected along the streamlines (gas molecule jumps) may explain this anomalously fast diffusion. At the end of this stage, the flame becomes fully developed and its height reaches the maximum $l_{\max}$. This phase is followed by a decrease of flame height, mass-loss rate, temperature and gas velocity over a relatively long period, characterising the decay phase. As combustion occurs mainly in the bulk of the fuel bed in this phase, the anomalously slow relaxation of the burning vegetation (see Fig. 2b and Table 1) corresponds to sub-diffusion ($\alpha < 1$) of gas particles that may be due to the slowing down of their propagation through the porous fuel (they are ‘trapped’ by upper layers of the fuel bed for a finite time), and through the dense air/flammable gas mixture owing to the buoyant gases. Once there is no longer sufficient chemical energy to sustain the flame, it extinguishes and only smouldering combustion propagating slowly in the condensed phase subsists (Ohlemiller 2002; Rein 2009). The transition from flaming to smouldering combustion begins already in the decay phase, when there is an insufficient amount of oxygen in the bulk to allow the initiation of a flame.

As the thermocouple and Pitot tube are placed at the same position with respect to the initial fuel bed surface,

the temperature and the vertical motion of gas particles are measured at different relative positions of the flame structure during its temporal evolution. Hence, the flame temperature and gas velocity reach their maximum values when the flame is fully developed because they are measured at the closest position to the continuous zone (see Fig. 3a where $l_{\max} > 1$ m is much higher than the thermocouple position for all the fuels studied). As $l < 1$ m in the beginning of the growth phase and the last period of the decay phase, the temperatures and velocities measured in these periods are those of the thermal plume and not the flame itself. Therefore, the convection coefficient, which is related to air velocity, varies with the vertical position in the flame.

There is a time delay Δt between the time at which the burning rate is maximum ($t_{\max}$) and that at which the flame height is maximum ($l_{\max}$), and it has been found to depend linearly on the fuel moisture content (Sun *et al.* 2006). The moisture content may not be the only cause of this delay because even for completely dried fuels, Δt is not expected to vanish because a time delay is required for the emitted gas molecules to diffuse and advect through the porous fuel bed before contributing to the flame. An experimental investigation of moisture content effect on this time shift is necessary. In the current work, the moisture content of the samples was ~8–10% for all experiments, Δt seems to not change significantly (within statistical errors) for the dead fuels considered (apart from a slight increase for straw), as shown in Table 2 where the maximal values of the flame geometrical and physical characteristics are summarised. The maximum flame height and mass-loss rate are largest for shrubs and lowest for straw. The mass-loss rate of the fuel bed is related to its heat release rate $\dot{Q}_f$ and its effective heat of combustion ΔH_c as $\dot{Q}_f = \Delta H_c \dot{M}_f$. These parameters, given in Table 2, depend on the geometrical and physico-chemical properties of the fuel particles and also on the

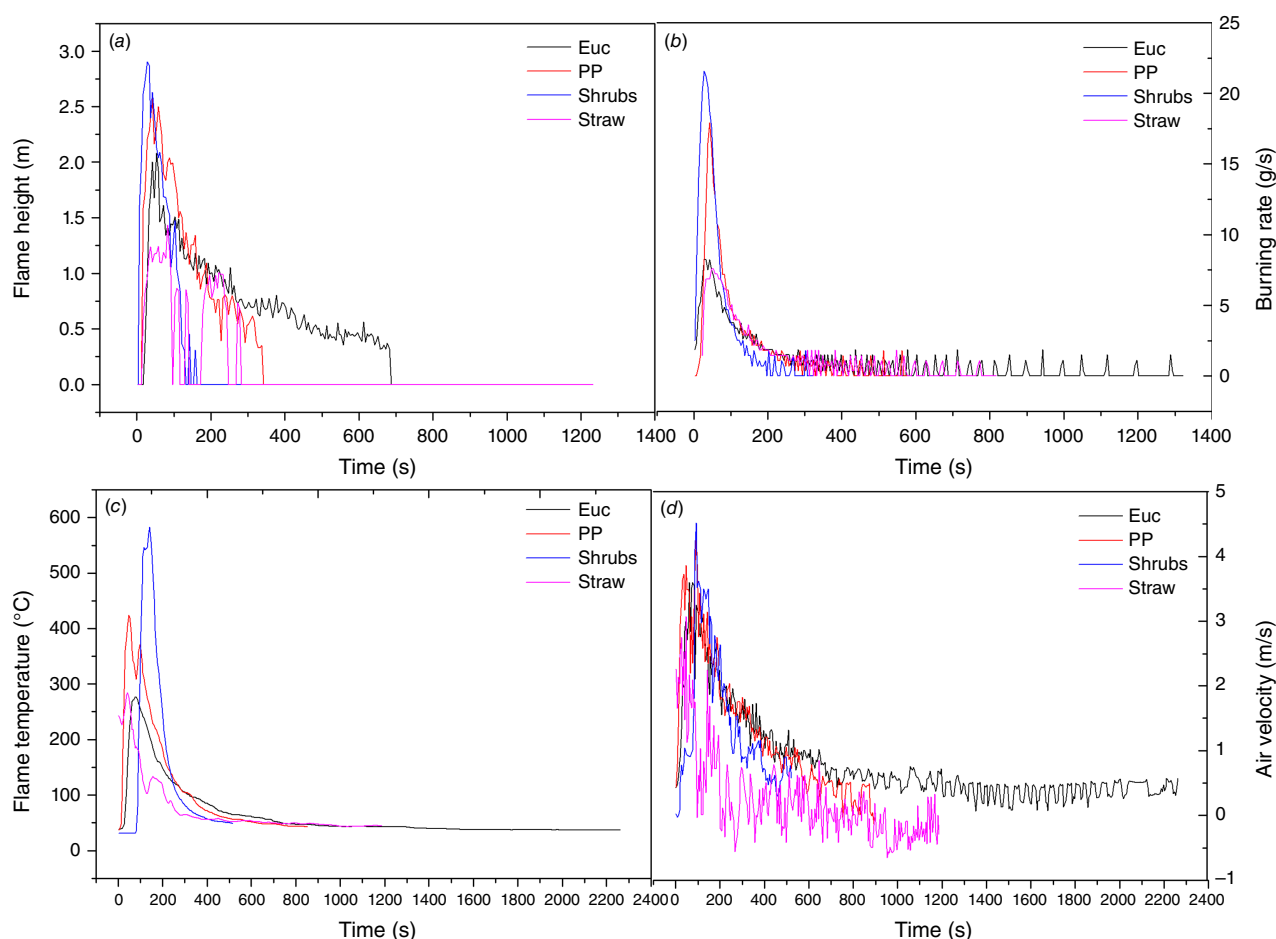


Fig. 3. Temporal evolution of: (a) flame height (m), (b) mass loss rate (g/s), (c) flame temperature (°C), (d) air velocity (m/s).

Table 2. Maximum values of some burning characteristics of several Mediterranean forest fuels.

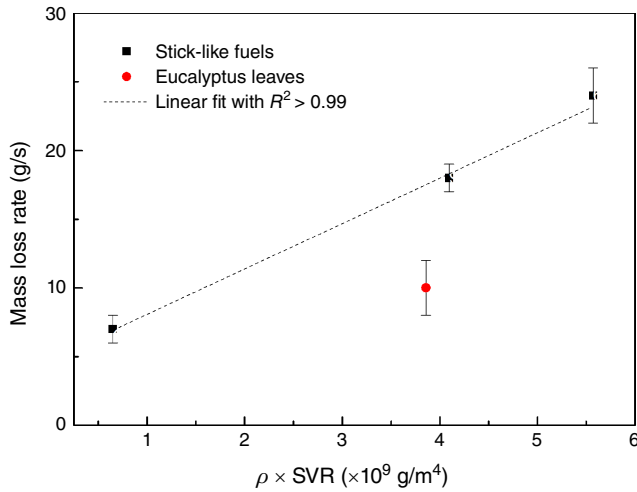
Fuel type	ΔH_c (kJ/g) (Pinto et al. 2017)	l_{max} (m)	$\dot{M}_{fmax}$ (g/s)	$\dot{Q}_{fmax}^*$	T_{max} (°C)	v_{max} (m/s)	t_{res} (s)	Δt (s)
<i>Eucalyptus</i>	22.5	1.9(3)	10(2)	1.1(1)	292(30)	3.9(4)	845(260)	7(6)
<i>Pinus Pinaster</i>	13.5	2.5(3)	18(1)	1.2(1)	397(19)	4.2(2)	242(83)	8(8)
Shrubs	16.9	2.9(1)	24(2)	2.0(2)	488(78)	4.9(4)	148(35)	13(8)
Straw	11.7	1.5(3)	7(1)	0.4(1)	334(34)	3.3(4)	296(67)	25(7)

experimental conditions. On one hand, the gross heat of combustion depends only on the vegetation species (Rivera et al. 2012), i.e. on their organic components involved in combustion reactions (volatiles and heavy organic compounds). On the other hand, the mass-loss rate corresponds to the rate of gas emission from the fuel particles, and thus depends on their surface-to-volume ratio (SVR) and density ρ_p (see Table 3 where their values as well as those of packing ratios are summarised). Assuming that the flammable volatiles are emitted through the whole particle surface, the rate of their emission is higher for plants with higher SVR. Moreover, for the same volume,

larger particle densities imply larger volatiles mass emission. Therefore, at first look, one can suggest the following linear relation $\dot{M}_{fmax} \propto (SVR \times \rho_p)$. From Fig. 4, it is clear that the maximum mass-loss rate increases with $(SVR \times \rho_p)$, but the linear trend seems to be observed only for stick-like fuels (*Eucalyptus* leaves deviate from this trend). Among the various reasons for this deviation, the following seem important: (i) the strong variations in the experimental data of SVR and ρ_p ; (ii) *Eucalyptus* leaves seem more closely packed than the other fuels. The packing ratio may also influence the mass-loss rate, because it contributes to trapping of the gases emitted from the bulk; (iii) not all the fuel particles

Table 3. Physical and geometrical properties of the fuel particles.

Fuel	Particle density (kg/m ³)	SVR (m ⁻¹)	Packing ratio (%)
Shrubs (Fernandes and Rego 1998)	780	6330–7950	3
<i>Pinus Pinaster</i> (Lamorlette et al. 2015)	660–927	4820–5500	2.5–3.5
<i>Eucalyptus</i> (Fernandes and Rego 1998)	650	5690–6180	3.6
Straw (Adapa et al. 2009)	268–286	2342	8.1–8.7

**Fig. 4.** Mass-loss rate vs $\rho \times \text{SVR}$ for the fuels considered in this work (the data are taken from Tables 2 and 3).

components are combustible because they are also composed of minerals, for instance; (iv) the pore distribution in leaves is different from that in needles (see Sabi et al. (2021) and references therein).

The mass-loss rate at ignition (for cone calorimeter tests) was found by Bouchiche et al. (2022) to be independent of the packing ratio (defined as the ratio of the bulk and particle densities). However, an optimum packing ratio for the rate of spread has been observed (Rothermel 1972). Hence, the packing ratio is expected to influence particularly the combustion dynamics of the fuel bed (time evolution of the mass-loss rate) because the diffusion of flammable gases through the porous fuel depends on the mean free path between the fuel particles. Therefore, the dynamic exponent β_2 is expected to vary with the fuel bed's porosity because combustion occurs mainly in the bulk during the decay phase of the flame, as mentioned above. Therefore, it is interesting to investigate the effect of the packing ratio on the burning rate dynamics for each vegetation species separately.

The time-dependent heat release rate (mass-loss rate) can be deduced directly from the time derivative of KWW Eqn 3:

$$\dot{Q}_f(t) = \frac{\Delta H_c (M_0 - M_{fmin}) \beta}{\tau} \left(\frac{t}{\tau} \right)^{\beta-1} e^{-\left(\frac{t}{\tau}\right)^\beta} \quad (6)$$

with $\frac{\Delta H_c (M_0 - M_{fmin}) \beta}{\tau}$ a constant that depends on the relaxation properties of the fuel. This law taking into account the different phases of flaming combustion can render the semi-physical models more realistic by replacing the constant heat release rate used in these models. It also improves the existing physical (CFD) models that use equations accounting only for normal diffusion/relaxation processes as discussed above.

Zukoski (1975) introduced a dimensionless heat release rate defined as the square root of the Froude number:

$$\dot{Q}_f^* = \frac{\dot{Q}_f}{\rho_a T_a c_{pa} d_c^2 \sqrt{g d_c}} \quad (7)$$

where d_c is the container diameter, and g , ρ_a , T_a and c_{pa} are the gravitational acceleration, density, temperature and specific heat of ambient air respectively. The variation of the maximum normalised flame height (l_{max}/d_c) as a function of $\dot{Q}_{fmax}^*$ allows the identification of three regions of the scaling behaviour of the flame: the buoyancy-driven diffusion flames ($\dot{Q}_{fmax}^* \leq 100$), the momentum-controlled jet fires ($\dot{Q}_{fmax}^* \gg 100$) and a transition zone between these (Thomas 1963; Zukoski 1986; Quintiere and Grove 1998; Drysdale 2011; Finney and McAllister 2011). In the case of buoyancy-controlled diffusion flames, the normalised maximum flame height increases as a power-law with $\dot{Q}_{fmax}^*$, the power exponent being: 2 for $\dot{Q}_{fmax}^* < 0.1$, 2/3 for $0.1 < \dot{Q}_{fmax}^* \leq 1$, and 2/5 for $1 < \dot{Q}_{fmax}^* \leq 100$, where the flame height becomes independent of d_c .

A two-fifth power law was observed for *Pinus Pinaster* needles and excelsior (Dupuy et al. 2003), chaparral fuels (Sun et al. 2006) and shrubs (Pinto et al. 2017). The results presented in Table 2 suggest that, on one hand, the dimensionless heat release rate is in the range $1 < \dot{Q}_{fmax}^* < 100$ for shrubs (in agreement with the results of Pinto et al. 2017), *Pinus Pinaster* (in agreement with the results of Dupuy et al. 2003) and *Eucalyptus*. On the other hand, $0.1 < \dot{Q}_{fmax}^* \leq 1$ for straw, which is consistent with the results found by Sahila et al. (2023) where $l_{max}/d_c \sim (\dot{Q}_{fmax}^*)^{2/3}$. This suggests that, for the same load and bulk density, the fuel type affects the scaling behaviour of the buoyancy-driven turbulent diffusion flames. A complete scaling analysis of the burning behaviour of these species is then necessary for the same load and bulk density to check this scaling exponent.

The flaming combustion duration defines the residence time of the flame t_{res} . In Table 2, the maximum flame height (and maximum mass-loss rate) decreases as the residence time t_{res} increases for the fuels considered with a saturation tendency for *Eucalyptus* and straw. This is in qualitative agreement with the experimental results found in the case of fire whirls (Pinto et al. 2017). Note the very large residence time of *Eucalyptus*, which is also characterised by a large heat of combustion.

Conclusion

An experimental study of the burning characteristics of several dead forest fuels was carried out. Samples of dead *Eucalyptus*, *Pinus Pinaster*, straw and shrubs with the same fuel load and bulk density were placed in cylindrical baskets and ignited from the bottom. The time evolution of the combustion parameters was found to follow the same trend for all the fuels considered. In the growth phase of the flame, the fuel mass-loss rate, flame height, temperature and upward gas velocity increase rapidly. This phase is characterised by an anomalously fast relaxation of the fuel mass ($\beta_1 > 1$) and corresponds to a super-diffusion of gas particles through the fuel bed surface caused by advection-induced particle jumps. Afterwards, a crossover (transition zone) appears when the flame is fully developed, and the burning rate attains its maximal value after a time that was found to be the same for all the fuels considered. Finally, these parameters decrease over a relatively long period (decay phase) characterised by an anomalously slow relaxation of the fuel mass ($\beta_2 < 1$) corresponding to a sub-diffusion of gas particles due to the trapping effect of the fuel bed. The first relaxation exponent β_1 is highest for *Pinus Pinaster* and lowest for straw, whereas the second (β_2) seems to be independent of the vegetation type. The relaxation times (τ_1 and τ_2) are shorter for shrubs than for *Pinus Pinaster* needles, *Eucalyptus* leaves and straw. The maximum flame height and mass loss rate are largest for shrubs and lowest for straw, which is due to their linear increase with particle density and surface volume ratio, as observed for stick-like fuels. The comparison between the Froude numbers (estimated from the maximum burning rate) of all the fuels considered suggests that the scaling behaviour of straw may be different from that of the other species. It will be interesting to investigate the scaling behaviour of the maximum burning characteristics of these species by varying the basket diameter, and to analyse the influence of their bulk and particle densities as well as their SVR on their combustion and relaxation properties.

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Author affiliations

^AFaculty of Physics, LEPM (Laboratoire d'Etude Physique des Matériaux), Université des Sciences et de la Technologie d'Oran Mohamed Boudiaf, BP 1505 El Mnaouer, 31000, Oran, Algeria.

^BDepartment of Mechanical Engineering, ADAI (Associação para o Desenvolvimento da Aerodinâmica Industrial), University of Coimbra, Rua Luís Reis Santos, Pólo II, 3030-788 Coimbra, Portugal.